

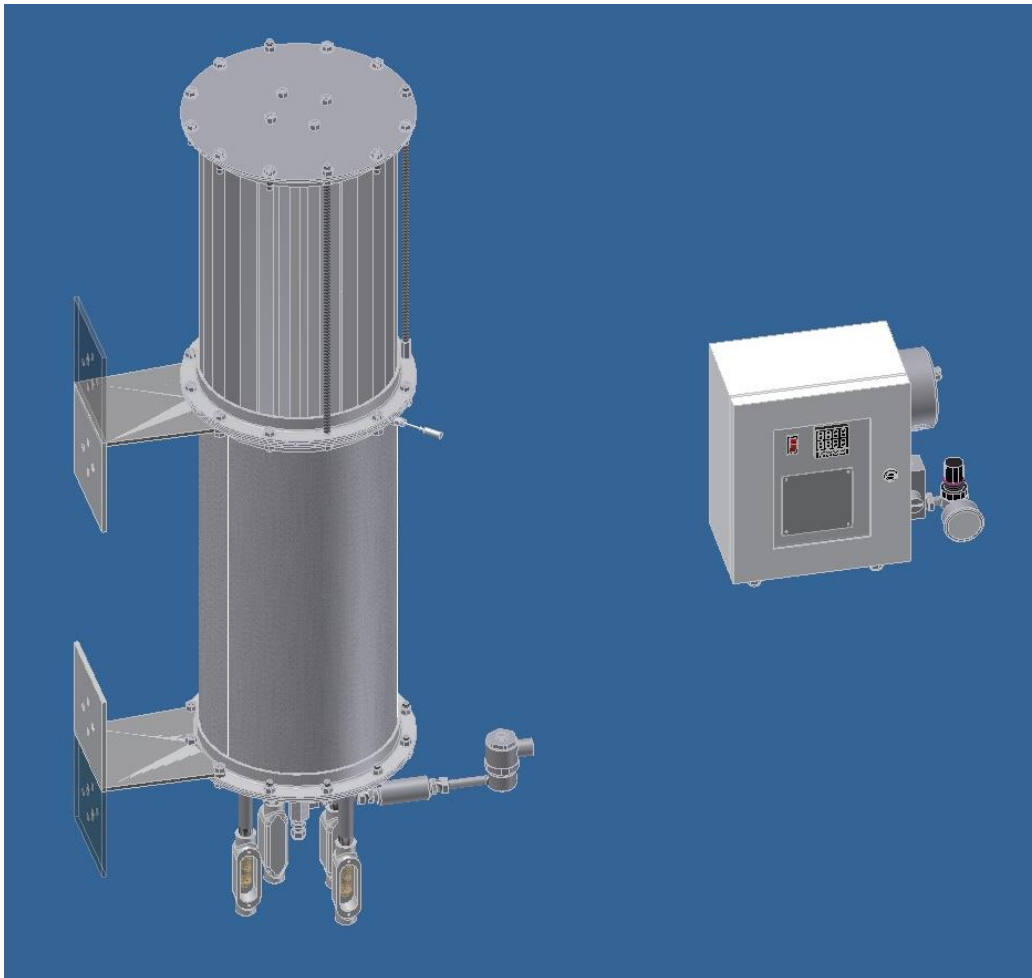
# Emissions Rx™ - Thermal Oxidizer

## Operations & Maintenance Manual

Thermal Oxidizer Model :

Serial Number :

Job Name:



# Operators Manual

## Index

Job Name:

Customer:

Order No.:

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## GENERAL

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The installation of Emission Rx T.O. must be done according to the applicable local and national codes.

The person, or persons, performing the work must be familiar with, and operate within, these applicable codes.

Installation, modification or service of the equipment supplied by Emissions Rx, must only be carried out by qualified, licensed persons.

Emission Rx. will not be held responsible for any damages caused by unqualified persons, unsafe practices or unauthorized modifications to the received goods.

The pressure drop in the gas line should be kept to a minimum. Therefore, the gas supply line should be as short as possible.

A complete Emission Rx. Start-Up Report must be completed by the installer and a copy sent to Emission Rx. This is a warranty requirement!

It is the responsibility of the qualified installation person to determine the following from the local gas authority (company) prior to starting the burner:

The installation of Emissions Rx Thermal Oxidizers must be done according to the applicable local and national codes. The person, or persons, performing the work must be familiar with, and operate within, these applicable codes.

Installation, modification or service of the equipment supplied by Emissions Rx, must only be carried out by qualified, licensed persons. Emissions Rx will not be held responsible for any damages caused by unqualified persons or unsafe practices.

The pressure drop in the Waste gas line should be kept to a minimum. Therefore, the Waste gas supply line should be as short as possible. A complete Emissions Rx Start-Up Report must be completed by the installer and a copy sent to Emissions Rx. This is a warranty requirement!

It is the responsibility of the purchaser to determine the following prior to starting the Thermal Oxidizer:

- Type of Waste gas
- Higher Heating Value (BTU per cubic foot)
- Maximum CO<sub>2</sub> (Carbon Dioxide) content of flue Waste gas
- Incoming Waste gas pressure
- Whether, or not, the Waste gas meter is pressure and/or temperature compensated.

Acquire the correction factor in order to determine the correct Thermal Oxidizer input Chemical composition of special Waste gases (Digester, Sewage, Manufactured etc.)

## 1.1 WARNING

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Read the following warnings and instructions before attempting startup of any Combustion Control device.

If you smell Waste gas: - Ventilate Room!

- Extinguish any open flame!
- Do NOT smoke!
- Do NOT tamper with or touch anything ELECTRICAL!
- Do NOT attempt to start the Thermal Oxidizer!
- Call the Waste gas supply company!
  - The installation, Start Up, Operation and Servicing of an Emission Rx Thermal Oxidizer must be done according to the manufacturers' instructions and applicable local and national codes.
  - Incorrect procedures can lead to the exposure of components of the fuel and combustion products which can cause serious bodily harm or death.
  - The storage of flammable liquids in an open container near the Thermal Oxidizer application, or in the same room, can be dangerous and should be avoided.
  - Before commencing any service work:
    1. Close off all fuel lines!
    2. Switch off and isolate electrical power to the equipment!

The use of Teflon tape as an oil and Waste gas pipe sealant can cause fuel valves to fail and create hazardous situations. If Teflon tape is used, ALL WARRANTIES ARE INVALID and Emissions Rx will not accept responsibility for any liability.

As a safety precaution, Emissions Rx has adopted the policy that all automated waste gas service piping must have 1 safety valve and a Back Flash Arrestor. Emissions Rx cannot accept responsibility for any installation which compromises this safety policy, even if local code states otherwise. **NOT FOLLOWING THIS Emissions Rx POLICY WILL RENDER ANY WARRANTY VOID!**

Any work performed on the Thermal Oxidizer, applicable interfaced equipment, or controls must be done by qualified and trained personnel familiar with the product and licensed to undertake such work. All work must be completed within the parameters of local and national codes.

Do not set up the flame on Emissions Rx Thermal Oxidizers without accurate flue Waste gas measuring instruments.

Correctly calibrated combustion instruments are the only reliable means to determine a safe operation of the Thermal Oxidizer.

All safety and control devices must be checked for correct operation and adjusted with the correct settings before operating the unit.

*This manual may be updated without notice. Consult your authorized distributor or Combustion Systems Co Inc. for the latest version.*

## 1.2 Checklist

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In order to avoid damage to the equipment, check the following prior to starting the Thermal Oxidizer:

- Heat Exchanger is installed properly (if applicable)
- Limits and operating controls on the Thermal Oxidizer are mounted and wired correctly (if applicable)
- Type and range of all controls are correct for the application
- Oxidation chamber dimensions are adequate for the required waste gas disposal rate
- Oxidation chamber is unobstructed to allow for required flow rate
- Thermal Oxidizer/Heat Exchanger can be fired, at the required firing rate, long enough to take flue Waste gas readings
- Thermal Oxidizer/Controls are installed correctly (if applicable)
- Thermal Oxidizer is installed correctly
- No damage to the Thermal Oxidizer has resulted from transportation or installation
- The electrical supply is correct for the Thermal Oxidizer Elements and for the control voltage
- The field wiring to the Thermal Oxidizer and control panel (if applicable) is in accordance with the shop drawings provided
- All field wiring meets applicable local and national codes
- Sufficient combustion air is available for the Thermal Oxidizer to operate at the desired temperature and excess air.
- All equipment interfaced to the Thermal Oxidizer controls are correctly wired and are functioning properly
- Waste gas is identified prior to purchase and remains with range specified.
- The waste gas train is complete and complies to the applicable code
- No damage to the component parts of the waste gas train has resulted from transportation or installation (if applicable)

- The waste gas lines are the right size, have been installed correctly, are clear of dirt, have been purged and do not leak
- The incoming waste gas pressure is adequate for the installed waste gas train (if applicable) and required firing rate this per Instruction Manual!



# ● Emissions Rx™ - Operations Manual

## 1.3 – Site Conditions Data Sheet

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Job Name: \_\_\_\_\_ Date: \_\_\_\_\_

Customer Name: \_\_\_\_\_

Ship to Address \_\_\_\_\_

Customer P.O. \_\_\_\_\_

Customer Project File No. : \_\_\_\_\_

Quoted Model: \_\_\_\_\_

Site Elevation FASL \_\_\_\_\_ Feet

Waste Gas Pressure Min (inches W.C.): \_\_\_\_\_ Max (inches W.C.) \_\_\_\_\_

Waste Gas through put (Btu/hr): \_\_\_\_\_ HHV \_\_\_\_\_ LHV \_\_\_\_\_

Waste Gas Volume (ACFM) \_\_\_\_\_ (SCFM) \_\_\_\_\_ (LPM) \_\_\_\_\_

Waste Gas Temperature Degrees \_\_\_\_\_ °F \_\_\_\_\_ °C

Waste Gas Analysis by	Wt. %	Volume %
Carbon	_____	_____
Hydrogen	_____	_____
Nitrogen	_____	_____
Oxygen	_____	_____
Sulfur	_____	_____
Moisture	_____	_____
Ash	_____	_____
Other 1. _____	_____	_____
Other 2. _____	_____	_____
Other 3. _____	_____	_____

Common Name: \_\_\_\_\_

Available Electrical Power Voltage: \_\_\_\_\_ V/ \_\_\_\_\_ PH/ \_\_\_\_\_ Hz

Code Construction:  None  NFPA  C1D2BCD  Other

If Oher Specify \_\_\_\_\_.

## 1.4 Unit Performance Data Sheet

---

1000 Btu/SCF Natural Gas Basis of Design (Maximum Btu/hr Input) \_\_\_\_\_

Emission Rx Waste Gas Delivery requirements:

1. Waste Gas Connection Size: \_\_\_\_\_ NPT 316 SS
2. Waste Gas Static Pressure Required at ERx Connection: \_\_\_\_\_ ("W.C.)

Waste Gas Flash Protection:

1.  N/A
2.  Solenoid
3.  Back Flash Arrestor
4.  Solenoid & Back Flash Arrestor

Emission Rx Combustion Air Requirements:

1. Combustion Air Required: \_\_\_\_\_ SCFM
2. Combustion Air Static Pressure: \_\_\_\_\_ Inches W.C.

Type of Combustion Air Delivery:

Type of CA Control

- |   |                                    |
|---|------------------------------------|
| 1. <input type="checkbox"/> Natural Draft                 | <input type="checkbox"/> N/A       |
| 2. <input type="checkbox"/> Owner Supplied C.A.           | <input type="checkbox"/> On/Off    |
| 3. <input type="checkbox"/> ERx Supplied Forced Draft Fan | <input type="checkbox"/> Modulated |

F.D. Motor Amperage: \_\_\_\_\_ Amps

Combustion Air Flash Protection:

- N/A
- Combustion Air - Flash Attenuator
- Combustion Air Back Flash Arrestor
- Combustion Air Flame Arrestor

T.O. Exhaust Flash Protection:

- N/A
- Exhaust Gas - Flash Attenuator
- Exhaust Gas - Back Flash Arrestor
- Exhaust Gas - Flame Arrestor

Control Voltage Standard: 120/1/60

Other [ ]: \_\_\_\_\_ V/ \_\_\_\_\_ PH/ \_\_\_\_\_ Hz

Control Circuit Amperage: \_\_\_\_\_ Amps

Heating Element Voltage: Standard 120/1/60 [ ] Other [ ]: \_\_\_\_\_ V/ \_\_\_\_\_ PH/ \_\_\_\_\_ Hz

Heating Element Control:

1. Continuous
2. Intermittent
3. PID Set Point

# 1.5 Model Capacity

Unit Model is based on Unit Capacity in Btu/hr. input. Larger units available upon request.

Design Air Input to Waste Gas Ratio (based on) 100% concentration of Waste Gas HHV																								
Btu/CF 1000 Liter Per CF	Gas Methane 1	LPM	CFM	Btu/M	Btu/s/hr.	CF/H	Max Air CFH	Btu/CF	Gas Propane 1	LPM	CFM	Btu/M	Btu/s/hr.	CF/H	Max Air CFH	Btu/CF	Gas Nitrogen 0	LPM	CFM	Btu/M	Btu/H	CF/H	Max Air CFH	
																								28.3168

⊕ Designed Air Flow delivered is based on Btu/hr. input, not Waste Gas Flow Rate.

★ Emissions Rx Design Capacity is based on Natural Gas - CH4 - 1000 Btu/CF HHV

◆ If actual heat value drops to 0 as is the case of N2 (Nitrogen has "0" Heating Value) the air flow will remain sufficient to support the design Waste Gas max Btu/hr. input analysis.

## 1.6 Model Nomenclature

Select capacity and features below.

<b>Emission Rx Model Number Matrix</b>										
	ER <sub>x</sub>	-1	-2	-3	-4	-5	-6	-7	-8 - - - - -	
	ER <sub>x</sub>									
<b>1</b>	<b>Unit Capacity in Btu/hr. Input @100%Sat.</b>						<b>5</b>	<b>Exhaust Protection</b>		
	<b>A</b>	0-10K Btu/hr.						<b>A</b>	None	
	<b>B</b>	10K-15K Btu/hr.						<b>B</b>	Back Flash Preventer	
	<b>C</b>	15K-40K Btu/hr.						<b>C</b>	N/A	
	<b>D</b>	40K-80K Btu/hr.						<b>D</b>	N/A	
	<b>E</b>	80K-140K Btu/hr.						<b>E</b>	N/A	
	<b>F</b>	N/A					<b>F</b>	N/A		
<b>2</b>	<b>Combustion Air Delivery</b>						<b>6</b>	<b>Electronics</b>		
	<b>A</b>	Natural Draft						<b>A</b>	None	
	<b>B</b>	Blower						<b>B</b>	1 PID Loop Indicating controller w/ T/C	
	<b>C</b>	C1D2 Pump						<b>C</b>	2 PID Loop Indicating controller w/ T/C	
	<b>D</b>	Owner Supplied						<b>D</b>	3 PID Loop Indicating controller w/ T/C	
	<b>E</b>	N/A						<b>E</b>	PLC 3 loop BMS/CCS Modulated Air	
	<b>F</b>	N/A					<b>F</b>	PLC & Cloud Storage		
<b>3</b>	<b>Waste Gas Protection</b>						<b>7</b>	<b>Unit Construction</b>		
	<b>A</b>	None						<b>A</b>	None	
	<b>B</b>	Flame Arrestor FA						<b>B</b>	NFPA 85	
	<b>C</b>	Blocking Valve BV						<b>C</b>	C1D2	
	<b>D</b>	Flame Arrestor (FA) & Blocking Valve BV						<b>D</b>	N/A	
	<b>E</b>	N/A						<b>E</b>	N/A	
	<b>F</b>	N/A					<b>F</b>	N/A		
<b>4</b>	<b>Combustion Air Protection</b>						<b>8</b>	<b>Options</b>		
	<b>A</b>	None						<b>A</b>	None	
	<b>B</b>	Back Flash Preventer						<b>B</b>	Rain Cap	
	<b>C</b>	N/A						<b>C</b>	Alarm Horn	
	<b>D</b>	N/A						<b>D</b>	Waste Gas Flow Meter	
	<b>E</b>	N/A						<b>E</b>	Panel Purge Kit	
	<b>F</b>	N/A					<b>F</b>	Continued Below		
Options Continued:										
<b>F</b>	Direct Cloud Reporting Service									
<b>G</b>	Recommended Spare Parts Kit for this unit									
<b>H</b>	Exhaust Heat Exchanger/Flash Arrestor Weather Shield									
<b>I</b>	Combustion Air Flash Arrestor Weather Shield									
<b>J</b>	Spark Ignitor & Ignition Transformer									
<b>K</b>	Safety Cage									
<b>L</b>	H <sub>2</sub> S Kit									
<b>M</b>	Specify									
<b>N</b>	Specify									
<b>O</b>	Specify									

## 2. Operation

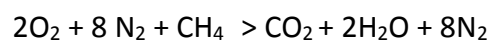
Thermal Oxidization (T.O.) has long been the standard method of destroying waste gas streams such as VOC's in large quantities. Traditionally T.O. devices have been installed in very passive environments which can tolerate extreme external temperatures such as, an open field, atop of structural towers in "non-hazardous environment". Natural gas combustion has been the preferred method of generating the heat necessary to allow oxidation to take place in these large systems. Excess air, internal to the waste gas stream or mechanically injected by a separate supply device, provides the Oxygen for Oxidation. With the advent of new regulations and tightening of existing allowable limits regulation such as Title VI of US EPA, and specifically the new TRI regulations, smaller and smaller quantities of VOC's emissions are being tolerated.

The waste gas stream, for this discussion relates to those compounds (natural or manmade) which are known as Toxic Release Inventory (TRI) gas or Volatile Organic Compounds (VOC's), that require oxidization by US EPA or other Jurisdictional air quality authorities.

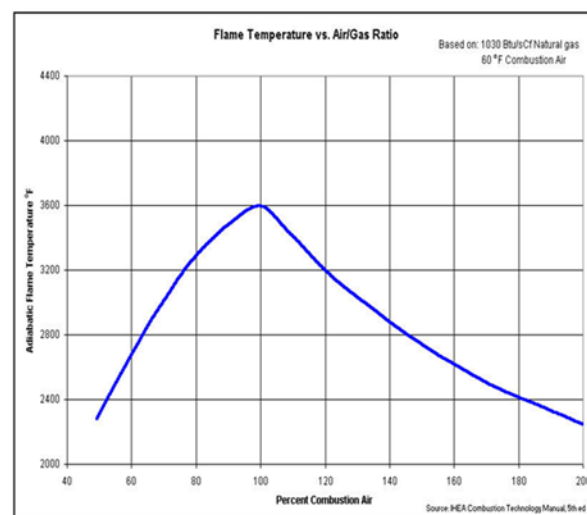
### 2.1 Theory of Operation:

TRI Gases or VOC's consist of Hydrogen and Carbon molecules which by themselves or in combination with other elements form highly toxic substances or are environmentally egregious in their natural state such as Natural Gas, which is mostly Methane (CH<sub>4</sub>). When Natural gas reacts with air in the

process of combustion heat and light is released. In that process air, containing 20% Oxygen and 80% Nitrogen is mixed with Natural Gas (Fuel) in a ratio of 10 Cubic feet of air and 1 cubic foot of fuel. Theoretically, there is enough Oxygen (O<sub>2</sub>) in 10 Cubic foot of air to completely unite with the Hydrogen and Carbon that contained in 1 Cubic foot of Natural Gas (CH<sub>4</sub>). If all of the Oxygen (O<sub>2</sub>) is completely united with all of the Carbon and Hydrogen then the reaction is as follows:

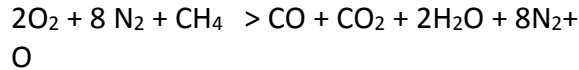


In the process Products of Combustion (P.O.C) are elevated to 3600 °F. This perfect mixture is said to be a Stoichiometric mixture.

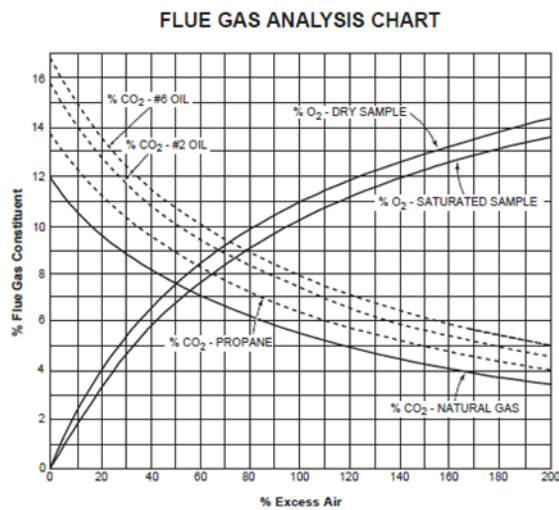
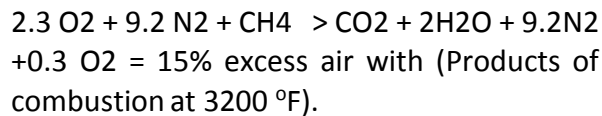


If perfect mixing of the fuel and air does take place it will result in forming Carbon dioxide (CO<sub>2</sub>) and Water Vapor (H<sub>2</sub>O) while releasing heat. Because the N<sub>2</sub> introduced with the air is inert, it may pass through the T.O. unchanged chemically but elevated in temperature.

If complete mixing doesn't occur then Carbon Monoxide (CO) is formed in various amounts as follows:



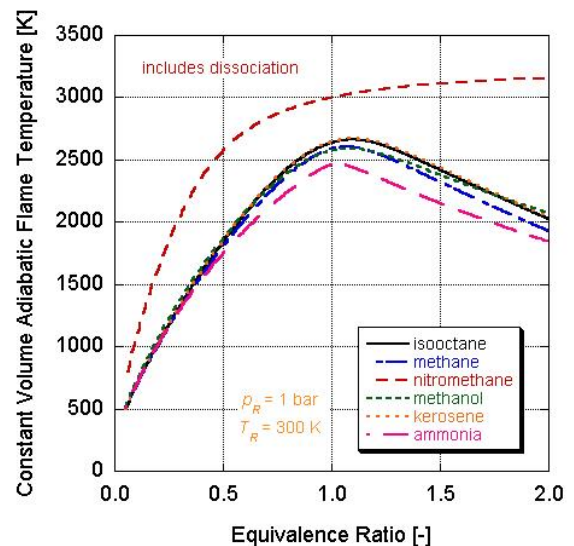
In order to ensure that more than enough Oxygen is available a minimum of 15% additional air molecules needs to be present during the process of combustion or 11.5 Cubic Feet of air per 1 Cubic foot of Natural Gas or:



The higher the excess air is, the lower the exit gas temperature of Products of Combustion (POC) will be. This is true for all gases listed in standard Tables of Combustion Constants (in Appendix B). If a large amounts of air is present in a Fuel/Air mixture, lowering the (Hydrocarbons/Air) ratio to (< 0.4:10) such that the amount of natural gas in the mixture is less than 4% of the total, it is said to be too lean for auto ignition or below the Lower Explosion Limit (LEL). This means that such

mixtures at ambient temperatures, when exposed to an Ignition source, will not react. On the other hand, if the mixture is too rich with fuel i.e. > 15% (1.5cf. fuel: 10 cf. air) it will not ignite because it is said to have exceeded the Upper Explosion Limit (UEL).

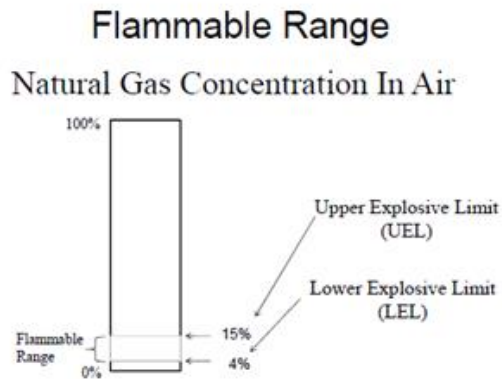
Complete conversion of hydrocarbons to H<sub>2</sub>O



and CO<sub>2</sub> through Thermal Oxidation can only take place if a sufficient amount of Oxygen (normally from Air) is available in the presents a of heat source for a sufficient amount of time for the reaction to take place.

The normal state of operation of many T.O. systems is that of a waste gas stream is in the LEL range with an insufficient quantity of hydrocarbons to allow for self-sustained combustion. Without an external heat source to heat the contaminated waste stream, typically a Natural gas flame to provide the heat source as stated earlier.

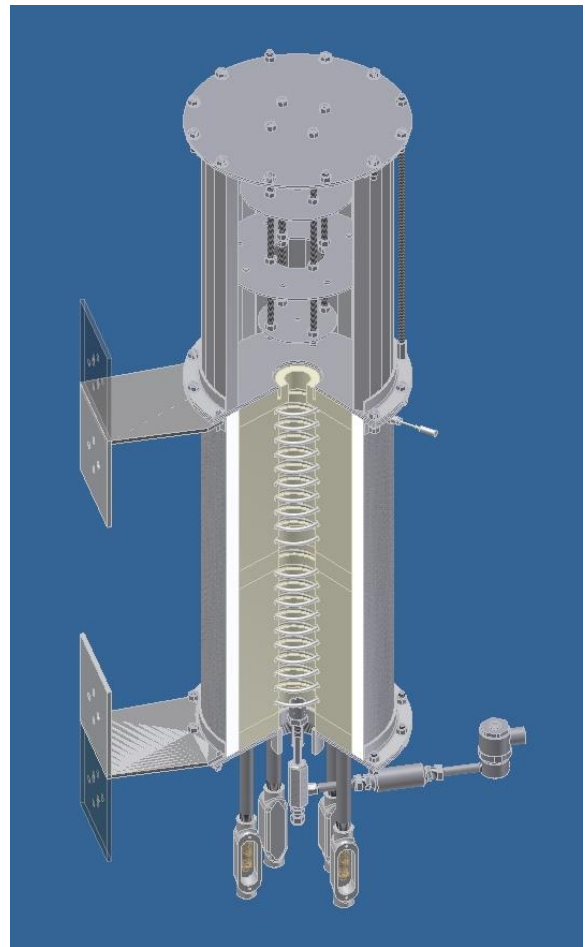
**2.2 Successful Destruction of VOC's** Excess O<sub>2</sub> and temperature are indicators as to quality of destruction.



Therefore depending on the waste gas to be destroyed the T.O. exit gas temperature is maintained between 1100 °F and 1800 °F and contains sufficient excess air to measure a minimum of 3% O<sub>2</sub> in the exit flue gas sample or 15% excess air in a chamber sized to provide for retention times of no less than 0.3 seconds on the low end to 2 seconds on gases containing Dioxins and Furans.

**2.3 Emission Rx.™** (Patent Pending) utilizes these principals on a smaller scale to meet existing and pending legislation for TRI gas regulations.

In its simplest form the Emission Rx.™ (Patent Pending) is an electrically heated thermal oxidizer designed to oxidize hydrocarbon waste gas streams. The Oxidization Chamber is preheated to 1500 °F by an electrical heating element, the waste gas is introduced in a vertical flow that must rise through the oxidation chamber and cross the heating elements. At the same time air is admitted at the unit's base and is mixed with the waste gas stream. Both the waste gas and the air are elevated to >1400 °F as they exit the oxidation chamber. Oxidation begins to occur at the point where heated fuel and air mixing begins. A secondary retention combustion section in series with the oxidation chamber provides sufficient



retention time to allow the hydrocarbons in the waste gas to react with the oxygen in the air to complete the combustion process which converts Hydrogen to H<sub>2</sub>O (Water Vapor) and Carbon to Carbon Dioxide CO<sub>2</sub>.

In practice the actual constituents being delivered to the Emissions Rx. T.O. unit are not known as they are often a mixture of several different sources and can vary from LEL to UEL within the waste gas stream or have no Hydrocarbons at all as in the case of a Nitrogen Purge. Because the waste gas stream may be an unknown analysis at any given time, its flow is limited such that, irrespective of the waste streams constituents, the Emission Rx. T.O. will deliver sufficient air to maintain a minimum of 11.5:1 ratio of air to waste gas stream.

Therefore the ERx T.O. can insure that the unit will always have sufficient Oxygen to oxidize the hydrocarbons in the waste gas stream.

This is accomplished by the introduction a fixed amount of air and limiting the waste gas stream to a customer prescribed design value and worst-case basis. Known gas characteristics are stated in the data sheet and becomes the basis of design to include Specific Gravity and Higher Heating Value. The waste gas flow is controlled by a flow limiting orifice and static supply pressure based on a stated gas.

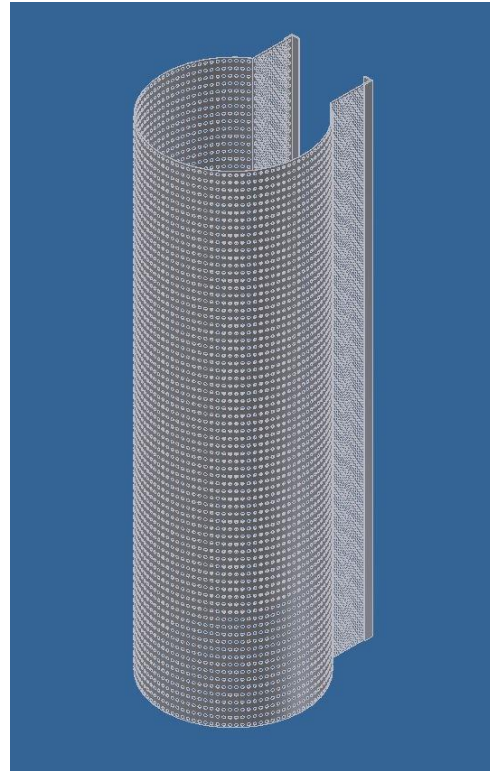
Heat in the T.O. cell is supplied by an electric heating element specifically design for this duty. Air may be supplied by simple induction from the bottom or by an air pump depending on the owner's requirements and destruction capacity. Oxidized Products of Combustion (POC), inert waste gas and associated heat is exhausted to atmosphere out the top.

The Emission Rx cell design allows for many iterations of this basic principal to include manually operated units which require minimal safety features in capacities of 10,000 Btu/hr or less, to units with capacities in excess of 80,000 Btu/hr installed in C1D2 area classification or constructed to NFPA 85, 75 and 68. These units may also be PLC based and contain contentious Emissions Monitoring which can both display or post compliance locally or to the owners DCS or directly to a cloud based data collection space supplied and monitored by Emissions Rx for maintenance and reporting purposes.

Options available with the Emission Rx unit.

Examples:

- Personnel Safety Cage
- H<sub>2</sub>S Kit



Optional safety cage shown above.



## 3.0 Installation

Your Emissions Rx Tri Gas Oxidizer comes assembled requiring only minor installation measures.

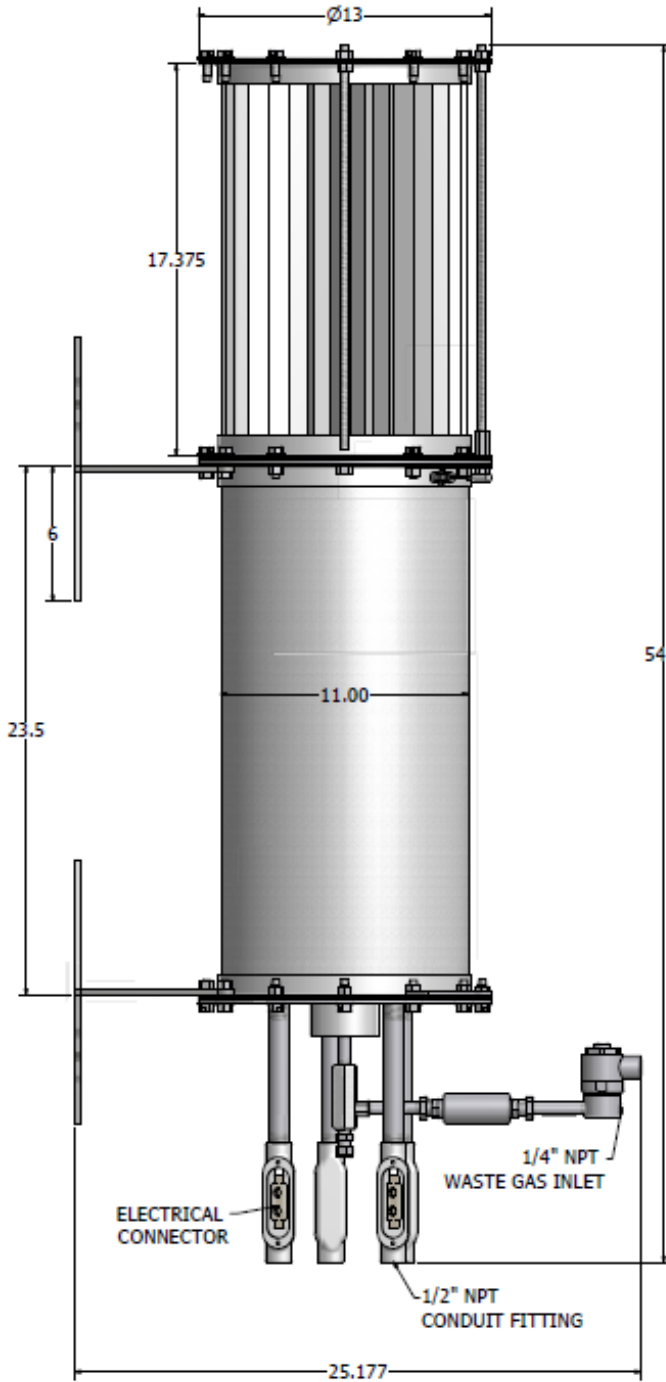
### **3.1 Contents:**

1. The Emissions Rx Thermal Oxidizer Assembly
2. Exhaust Gas Heat Exchanger
3. A universal mounting bracket (Shipped Loose). These brackets can be installed in many different configurations to meet your specific needs. (See recommended mounting arrangements). Also included is the Mixing Nozzle and Venture assembly shipped loose for field assembly.
4. Single PID Temperature control system w/ (1) T/C and CID2 Purge Panel kit installed.
5. Multi Source TRI Gas Manifold assembly containing:
  - a. TRI Gas Blocking Valve
  - b. TRI Gas Flash Arrestor
  - c. Optional pressure gages and flow meters.
  - d. Optional ½ inch Stainless Steel piping manifold with NPT fittings and Isolation Ball Valves (Maximum 10 source taps)
6. Depending on the model you have selected, additional components may also include:
  - a. CID2 Air Pump
  - b. Pressure Regulator
  - c. Personnel protection safety cage

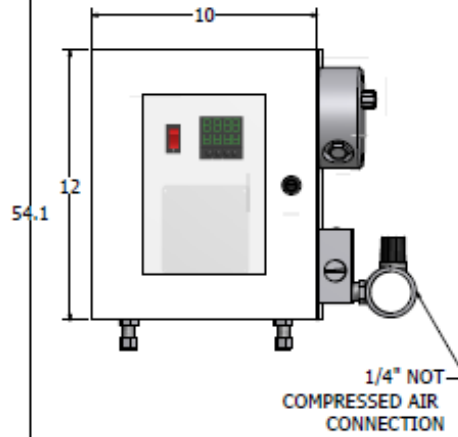
### **3.2 Location:**

- Identify a location for the Emissions Rx™ Thermal Oxidizer that is compatible with local codes and restrictions. (CID2 classification is certified on site).
- Proper Ventilation is required by code. The Oxidizer requires atmospheric air (NO concentrated oxygen) to operate and will emit CO<sub>2</sub> + N<sub>2</sub> + H<sub>2</sub>O + inerts in the sample at <300 °F.
- Make-up air may be required if installed in a walk-in enclosure.
- If the Emission Rx™ is located where personnel may come in contact with the oxidizer shell, the unit must be ordered with a Personnel Protection Safety Cage.
- Capacity and pricing calculations based on 10ft maximum distance between control panel and Emissions Rx unit.
- If utilizing air pump purchased with unit, air pump must be placed within 10ft of the control panel and be plumbed with a minimum of 3/8 inch stainless steel tubing.
- If utilizing air pump supplied by others, there must be a minimum of 20 PSI and 4 CFM at the point of connection of the control panel.

• **3.4 Dimensions**



Emissions Rx™	
CAPACITY	15K BTU/Hr
AMPERAGE	16 A
VOLTAGE	110 VAC
POWER CONSUMPTION	1.76kW
AIR PRESSURE	20 PSIG
AIR FLOW	4 CFM



Shipping Data	
Length	
Width	
Height	
Weight	
Number of containers	

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\*All measurements are in inches\*

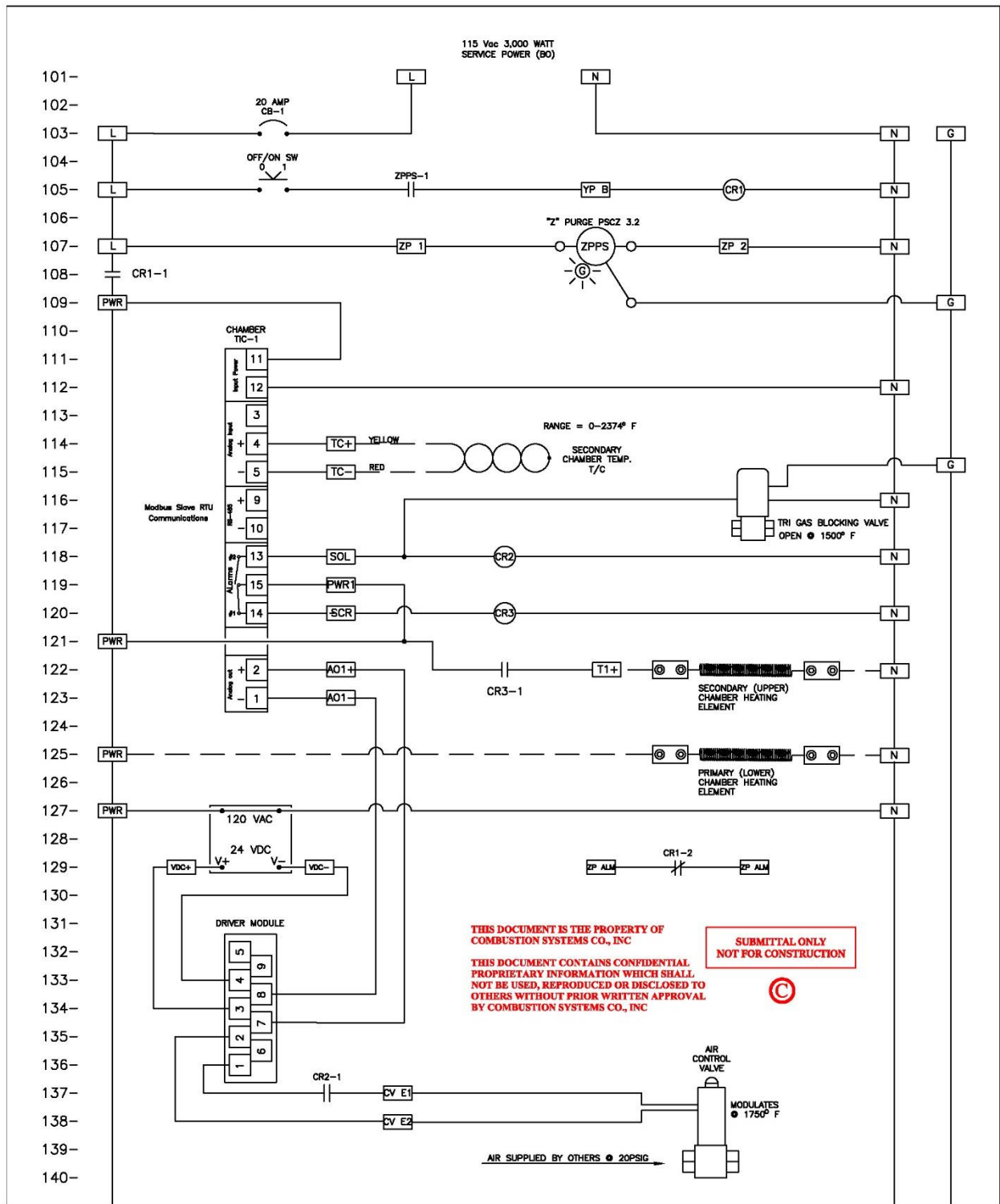
**3.5 By Others:**

The owner shall install an isolation valve upstream of Emissions Rx™ Tri gas point of connection. This valve shall be sized and constructed for the intended use.

# 4.0 Electrical

Each unit has been tested in the factory for proper operation and is ready for installation per the attached instructions. Power should not be applied to the system until the installation check is complete.

## 4.1 Wiring Diagram



### 4.1.1 Wiring Diagram Notes

- 1.) Power to the Pressurized Air Source is supplied and initiated (by the others).
- 2.) Unit Power is available when applied to L1 and L2.
- 3.) Control Power is present when ON/OFF Switch is in ON position however operation is inhibited until panel door is closed.
- 4.) When the panel pressure is sufficient to close the Purge Air Pressure switch ZPSS-1 the system will power up and start the heat-up sequence. (See Sequence Timeline).
- 5.) All Set point temperatures are permanently set and no adjustments are necessary.
- 6.) Initial warm-up. - The unit will reach 1500 F in about 4 hours as displayed on the digital indicator display.
- 7.) The Waste Gas Blocking Valve will open emitting gas in the Oxidation Chamber.
- 8.) Chamber temperature will be maintained at 1750 F once Gas is emitted into the Oxidizer Chamber.
- 9.) Should the Oxidization Chamber drop below 1500 F the Waste Gas Blocking Valve will close.

### 4.2 Power requirements:

Unless otherwise noted in the installation manual supplied with the unit, all units utilize 115 volt single phase power and must be have supply service protection, conduit and wiring properly sized for the location and power consumption of the system supplied.

- Each element will consume 1000 Watts (example unit in this packet utilizes 2 elements)
- Control panel will draw no more than 20FLA at 110 VAC.
- Oxidation air source, whether supplied with the order or by others, will require additional power and circuit protection not supported by the control panel. (See the individual wiring diagram for details)

### 4.3 Electrical Connections:



Your Emissions Rx T.O. only requires termination to (4) lugs located inside the ½” LC supplied with the unit. A 9/32” Hex head wrench is required to make the final connection. The top lug has been tightened to the heating element but should be checked for tightness. The lower lug is loose and ready to receive the field power connection. **Caution should be taken pulling field wire into the LC as the heating element lead is attached to a ceramic electrical isolation tube which is easily fractured.**



*¾” LC conduit containing field terminal for power wiring.*

#### **4.4 Electrical Checks**

After the unit has been mounted in accordance to **Section 3.0** and prior to landing supply wiring. Check all coil terminations (4) for continuity to the casing and frame. There should be no continuity from the elements to the casing or frame. If there is continuity, stop and contact your distributor. **DO NOT** apply power to the unit.

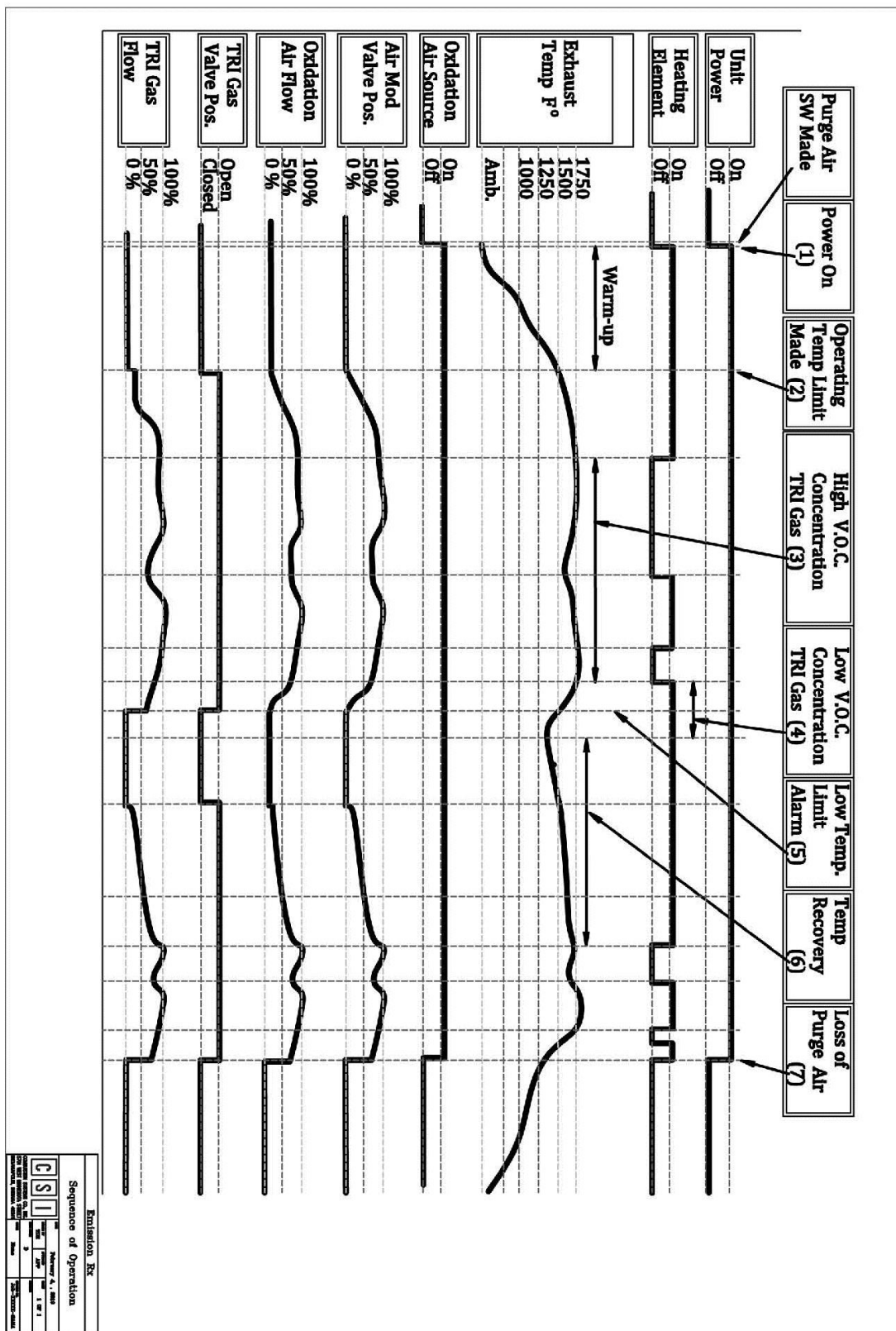
#### **4.5 Electrical Gas Tight Connections:**

Any junction or connection made to the control panel must be installed in conjunction with and area classifications where applicable.

High Temperature Silicone Caulk has been installed from the factory on heating element leads entering the field connecting LC conduit supplied with the unit. This serves to prevent gas from inside the cell from exiting the Cell.

# 5.0 Unit Operation

## 5.1 Emission Rx™ Event Reaction Chart

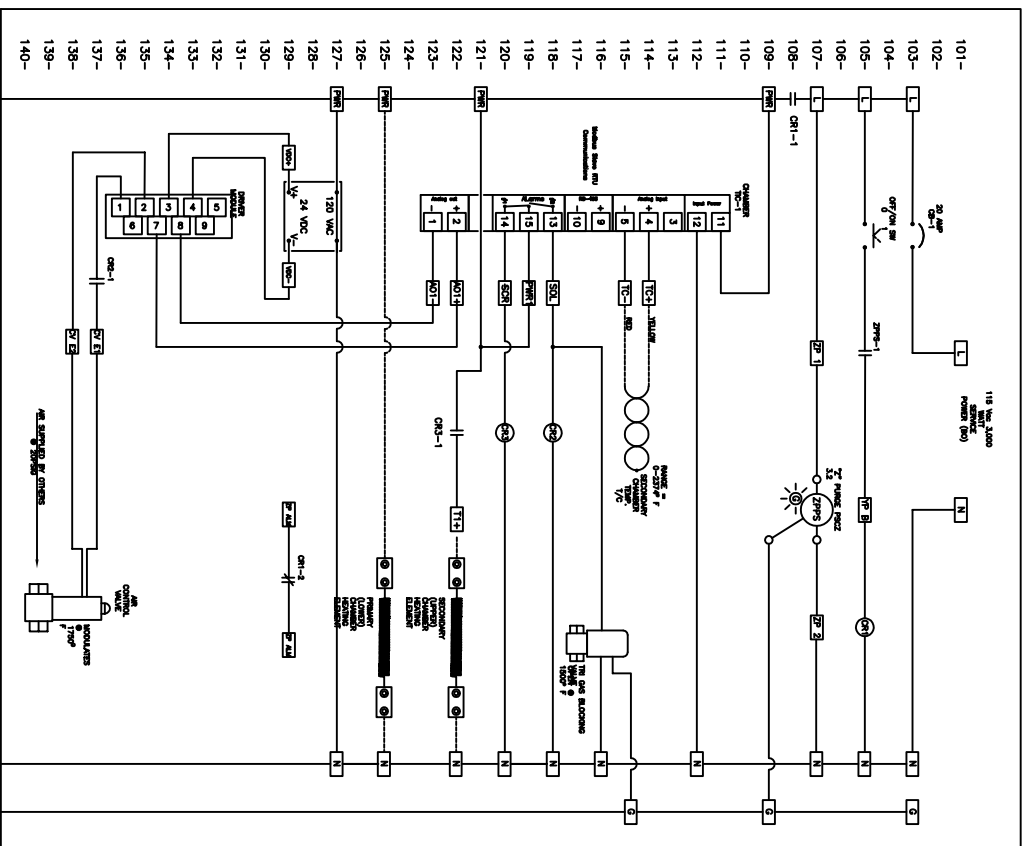


Emission Rx™ Event Reaction Chart

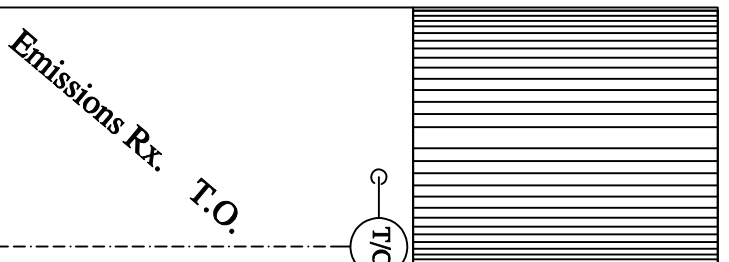
## 5.2 Emissions Rx™ Thermal Oxidizer

Model ERx -A-D-C-B-B-B-B-E contains an exhaust gas T/C and an Indicating Temperature Controller (ITC) which is designed to maintain a constant 1,750 °F at the Exhaust of the ERx T.O. by cycling the heating element power which will reduce electrical usage and extend heating element life. Per the Electrical Section 4.0. Power and field wiring between the ERx Control Panel and the Oxidizer unit is by the owner. The Unit ON/OFF switch is located inside the control panel and must be switched to the ON position before closing the enclosure door.

1. The unit will not actually apply power to the control circuitry nor Oxidizer cell until the enclosure is purged with fresh air delivered by the Combustion Air Source (e.g. : ADI pump).  
Once the Control Panel is pressurized, power is applied to the Temperature Indicating Controller (TIC) and the Cell will begin to heat per Column (1)- (2) in the above control sequence time line.
2. When the Cell temperature reaches its operating limit, the Oxidizing Air Valve (OAV) is released to modulate to maintain the set point temperature at 1,750 °F and the TRI Gas Blocking Valve (TRI-GBV) is opened. As the OAV opens TRI gas is induced into the Cell through the Venturi.
3. High Volatile Organic Compound (VOC) concentration in the waste gas will drive temperature in the cell higher and the Heating Element will be turned off allowing Oxidation of TRI Gas to drive the thermal process.
4. If the TRI Waste Gas concentration of VOC's diminishes or is replaced with Inserts such as N<sub>2</sub> or CO<sub>2</sub> the Heating Element will again be energized to bring the Oxidization Chamber temperature back to set point.
5. If however Cell temperature drops below Oxidization temperature the heating element will be powered, the OAV will close, and the TRI-GBV will close if the heating element cannot recover or maintain Oxidation Temperature.
6. Once the Cell Temperature recovers the system will return to operation mode 2 above.
7. Loss of Purge Air for any reason will shut the Unit down by killing power to the control system completely and close a set of alarm contacts which can imitate an Optional Alarm Horn or Light or Both.



\*See wiring diagram for sequence of operations

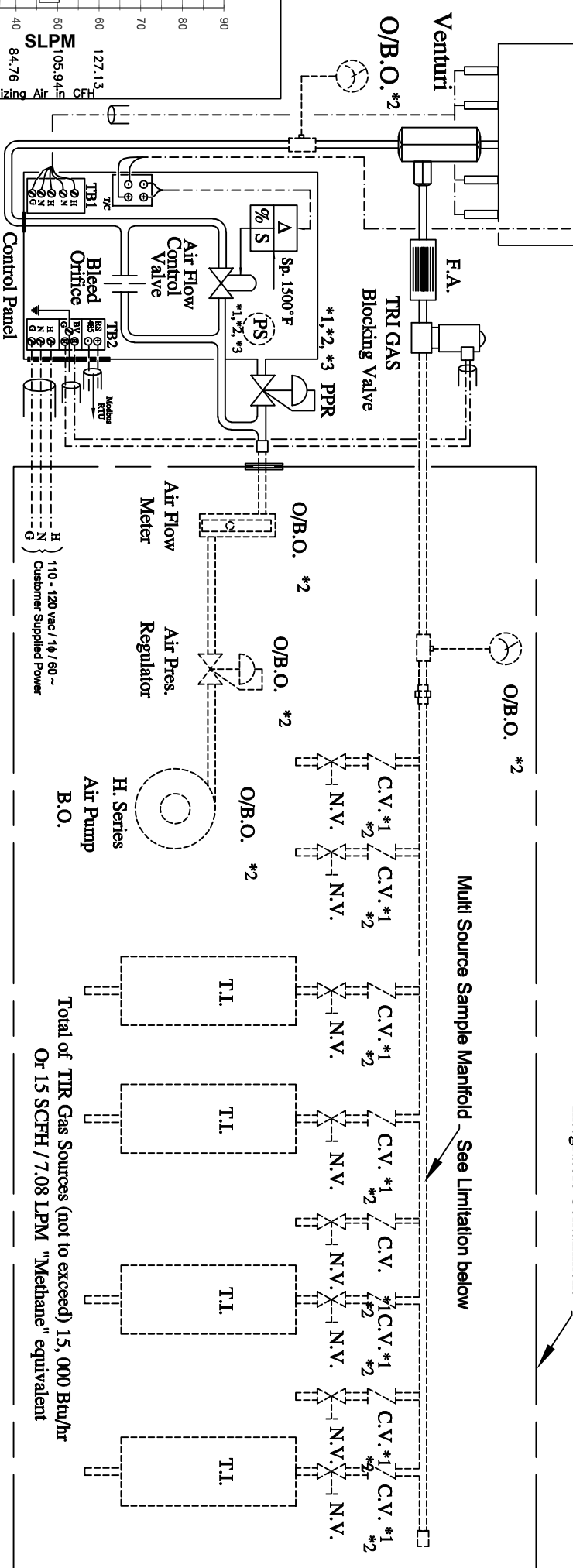


- Scenario#1 (4) Test units are delivering test samples with 0% (by v.) Combustibles 1.46LPM ea. for a total of 5.85 LPM.
1. Heating Elements are 1000 watts ea.
  2. (2) Heating Elements = 6,824 Btu's/hr
  3. 10% R&C Loss allows 6,141Btu's available
  4. 6,141Btu's will elevate 18.2lbs./hr. of air to 1400°F
  5. 18.2 lbs. = 241.7 SCFH of Air or 114.54 LPM of combined air and sample. This results in a sample feed rate of 7.43 LPM or 15.74CFH of Inert's such as Nitrogen
  6. If the sample is all Inert's such as Nitrogen the unit capacity is 7.43 LPM.

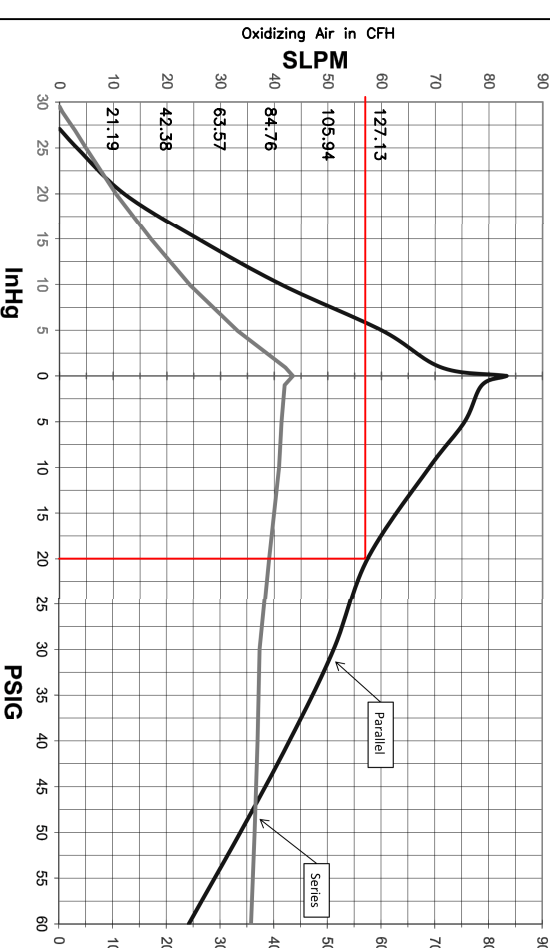
- Scenario#2 the (4) Test units are delivering test samples with 100% Combustibles (by v.) in the form of Natural gas at 1000 Btu/CF. at a rate of ~ 1.8575 LPM ea. or 15.74 CFH of Natural Gas total or 15,740 Btu/hr input.

- Scenario#3 the Cell is capable of combusting 17,000 Btu's/hr of Propane or Natural gas while maintaining 7% O2 and >20 PPM of CO @ > 1500°F and could require as many as (2) ADI Pumps to deliver sufficient air at max capacity.

9. A single ADI Pump per the attached is 57.5LPM @ the desired static or 121.84 CFH.
10. A single ADI Pump per the attached limits the units capacity to 55% of rating for both inert's and combustibles.
11. Purge air consumption will lower capacity as well, and needs to be deducted from these capacities.



### H-Series Dia-Vac - H302 (60 Hz)



Max Pressure = 60.0 PSIG (4.13 BarG) Parallel  
 Max Vacuum = 27.1 InHg (917 mBarG) Parallel  
 Max Flow = 83.3 SLPM (2.94 SCFM) Parallel  
 43.4 SLPM (1.53 SCFM) Series

\*Test results are averaged, and therefore should be considered approximate.  
 -These test results are for reference only, and are intended to help provide information to the user when determining which pump to buy. Actual pump performance will depend upon the user's application.  
 -For 50% duty cycle, reduce output by 17%.

#### Explanation of Abreviations

- B.O. By Others.
- C.V. Check Valve
- F.A. Flame Arrestor (Option)
- N.V. Needle Valve (Optional by Integrator)
- O/B.O. Option or by Others.
- PS. Pressure Switch (CID2 Required)
- Sp. Set Point
- T.B. Terminal Board
- T.I. Test Instrument
- V.B. Vent Breather (Option)

#### Symbols :

- Electrical Wiring
- Conduit
- (TIP) PID Loop
- Temp. Indicating Controller

#### Notes:

- \*1 Require for CID2 area classification
- \*2 Available upon request
- \*3 Purge Panel Regulator

**Multi Source Sample Manifold**  
 Patent Pending ©

Emissions Rx  
 Product Development

DATE: April 11, 2019

DRAWING NO. J-XXXX-XXXX

SHEET 2

REV	REVISION DESCRIPTION	BY	DATE
1	Added Modbus RTU 485 Terminals to Panel	TE	07/19/19
2	Corrected typo in Scenario #1 to 241.7 SCFH	DG	03/19
3			
4			

**CSI**

COMBUSTION SYSTEMS CO., INC.  
 5701 WEST MINNESOTA STREET  
 INDIANAPOLIS, INDIANA 46241

DATE: April 11, 2019

DRAWING NO. J-XXXX-XXXX

SHEET 2

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## 6.0 Maintenance

The Emissions Rx Thermal Oxidizer has been designed to minimize routine maintenance requirements and yet is fully repairable with original CSI supplied parts. Using components supplied by other than CSI will void all warranty, safety and performance guarantee.

The following table recommends component inspection and intervals for various devices external to the control panel.

Maintenance & PM Schedule					
Group	Item	Action	Test for:	Frequency	Range or method
External Piping & Components	Pipe Fittings	Tighten to Spec.	Leaks	Quarterly	Bubble Test
	Back Flow Flame Arrestor	Check for plugging	Flow at Static	Quarterly	Per Troubleshooting Guide
	Waste Gas Blocking Valve	Bench Test	Seal at Static	Annually	Per Troubleshooting Guide
	Mixing "T" Venturi	Remove & Inspect	Flow at Static	Annually	Per Troubleshooting Guide
	Nozzle	Remove & Inspect		Annually	No Erosion or Corrosion
Cell	Exhaust Heat Exchanger	Remove & Inspect	Carbon build up on interior	2000 hours of operation	Use 80 PSI Compressed Air to Puff from Outside to Inside
	Secondary Chamber Oxidizer Target Puck	Remove from Top & Inspect	Visible Deterioration	4000 hours of operation	No Erosion or Corrosion
	Secondary Chamber	Remove Jacket & blanket - Inspect	Visible Deterioration	4000 hours of operation	No Erosion or Corrosion
	Secondary Chamber Heating Element	Test in place	Test for open Circuit & Tighten Field Connectors	4000 hours of operation	Element Resistance >15 Ω & < ∞Ω
	Primary Chamber	Remove Jacket & blanket - Inspect	Visible Deterioration	4000 hours of operation	No Erosion or Corrosion
	Primary Chamber Heating Element	Test in place	Test for open Circuit & Tighten Field Connectors	4000 hours of operation	Element Resistance >15 Ω & < ∞Ω
Control Panel  No other Preventative maintenance required	Z-Purge	Open Panel Door	Z-Purge Operation	Monthly	Loss of Power to Controls
	By-Pass Orifice	Remove and clean	Blockage	2000 hours of operation	#74 Drill
	Any other failures inside the control panel should be return to CSI for Repairs.				
NOTES: Frequency of preventative maintenance is dependent on cleanliness of Compressed air (used for oxidation source) and Waste Gas Stream and is recommended based on commercial gas such as Natural Gas or Propane on the wetted components and Filtered Air. Actual Frequency may vary with the waste gas and compressed air source. Frequency may change from application to application. Maintenance personnel should make note of results of PM analysis before shortening or lengthen the recommended frequency.					

## 7.0 Troubleshooting Guide

Working on Electrical Equipment while powered implies a level of precaution which should be strictly observed and conducted by a qualified person. If after following this guide you cannot determine the probable cause of your problem contact your Emissions Rx Representative or call Combustion Systems Co., Inc. before attempting to disassemble the unit. Careless disassembly is potentially dangerous and can cause additional damage and cost to resolve.

### Forced Draft Units - NFPA & C1D2

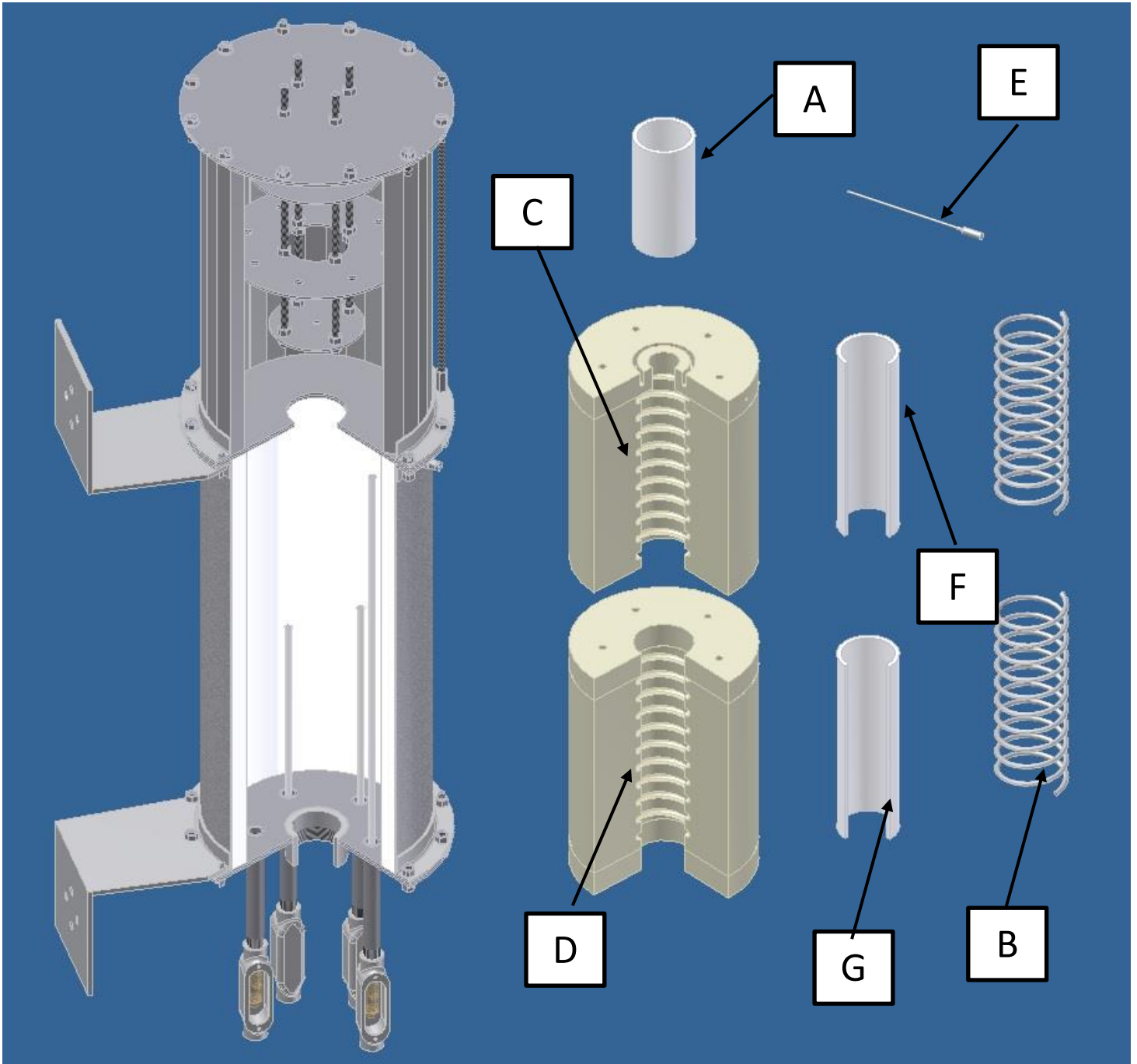
WARNING: Before troubleshooting, refer to sequence of operation timeline and general operating procedures for Emissions Rx™ Unit.

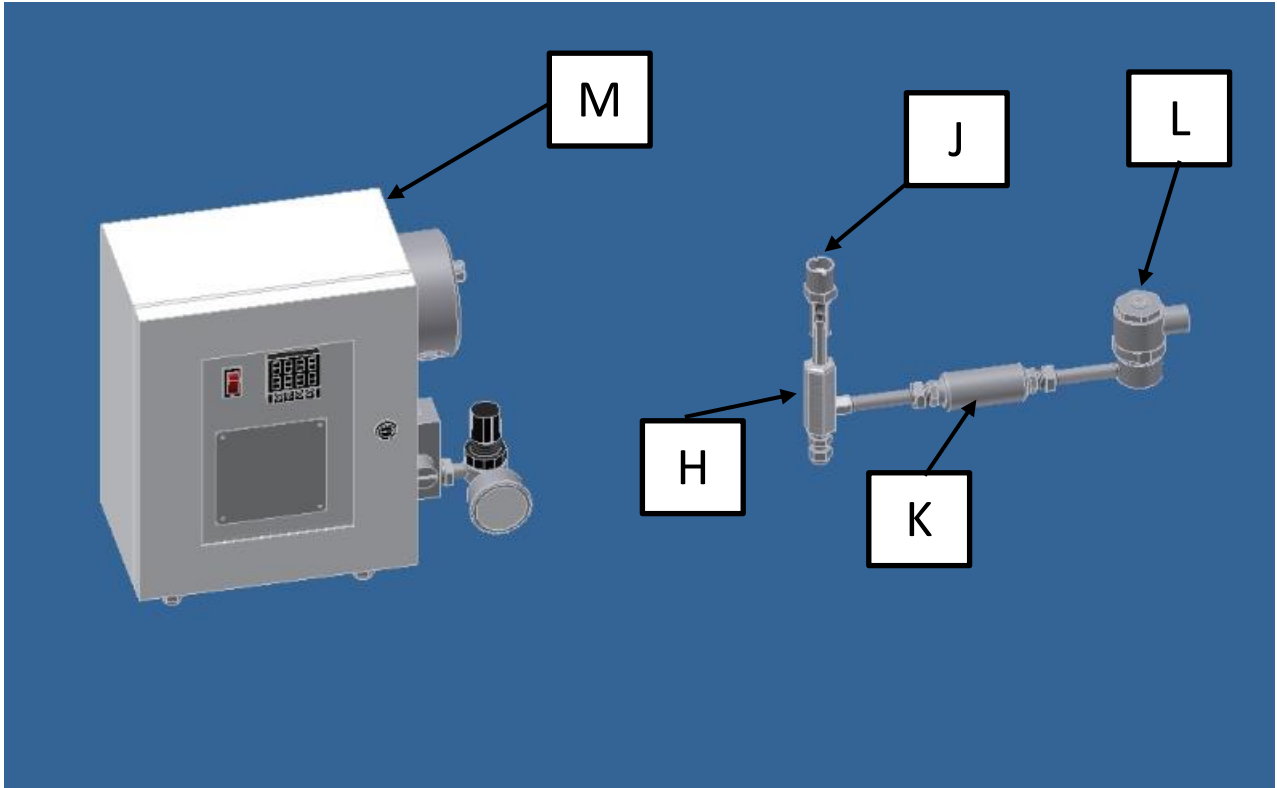
Problem	Error #	Probable Cause	Check	Appropriate Range	Actions
Unit not reaching operating temperature	1	No Power to Control Panel	Is power turned on to panel?	110VAC to 120 VAC	Apply Power to Unit
			Check if Fuses or Circuit Breaker is Blown		Reset circuit breaker/replace fuse
			Verify power to Z-purge Switch		Check wiring
			Is Z-Purge green light on when panel is shut?	Green light is lit	Verify proper air supply, verify properly sealed panel, Replace Z-purge switch
			Is controller powered ?	Visible display	Check CR1 for proper operation, bench test controller
			Is TIC calling for heat?	110 VAC to 120 VAC	Verify correct power at element connections
	2	Loose Heating Element Connection	Check for loose field connections at element leads	xxxx in-lbs	Tighten terminals to spec.
	3	No power to lower heating element	Check voltage across lower element field connection	110VAC to 120 VAC	Check field wiring
	4	Open lower heating element	Check amperage draw of element	7.9-8.9 amps	Replace lower heating element
	5	No power to upper heating element	Check for voltage across upper element field connections	110VAC to 120 VAC	Bench test zero cross over relay and single loop controller, Replace thermocouple
6	Open upper heating element	Check amperage draw of element	7.9-8.9 amps	Replace upper heating element	
7	High oxidation air flow	Check air supply pressure at panel connection	20 PSIG	Adjust pressure accordingly, verify bleed orifice installation	
8	Air modulation valve failed open	Check air pressure at mixing tee	Less than 20 oz static pressure	Replace modulation valve	
No Sample flow within operating temperature range	9	Blocking Valve not opening	Check voltage on solenoid	110VAC to 120 VAC	Bench test or replace controller,
			Bench test solenoid	20 scfh @ 10"WC static pressure	Replace blocking valve
	10	Flame arrestor plugged	Bench test flame arrestor	20 scfh @ 10"WC static pressure	Replace flame arrestor
	11	Failed thermocouple	Check controller for Temp reading	Type K thermocouple range	Replace thermocouple
12	No airflow at mixing tee	Check air leaks in piping	N/A	Return to manufacturer for repair	
Carbon Collecting Exhaust Heat Exchanger	13	Waste gas out of Specification	Analyze Waste and compare to Specification.	See Specification at purchase.	Notify the factory
	14	Waste Stream pressure too high	Check Waste Stream pressure	(3-4) " W.C.	Reduce pressure and Flow
	15	Low Combustion Air Flow from Blower	Check unit for blockage	20 PSIG at panel connection	Check for Proper Operation

## 8.0 Replacement Parts

### Emissions Rx™ Replacement Parts:

Model ERx -B-B-D-A-B-E-B-B & C shown below





Tag	Item	Part #	Qty.
A	Exhaust Liner	CSI-1101	1
B	Heating Element	CSI-1201	2
C	Retention Chamber	CSI-1301	1
D	Oxidation Chamber	CSI-1302	1
E	Thermocouple	CSI-1401	1
F	Retention Chamber H <sub>2</sub> S Liner	CSI-1102	1
G	Oxidation Chamber H <sub>2</sub> S Liner	CSI-1103	1
H	Mixing Venturi	CSI-2101	1
J	Oxidation Nozle	CSI-2102	1
K	Flame Arrestor	CSI-2103	1
L	TRI Waste Gas Blocking Valve	CSI-2104	1
M	Control Panel	CSI-3100	1

*The above parts list is a partial list of major components. Each unit is shipped with a complete parts list specific to your unit.*

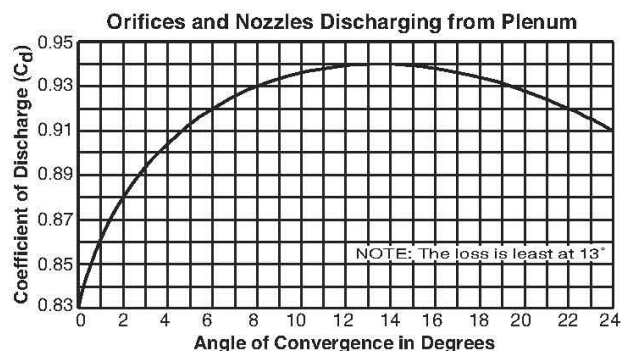
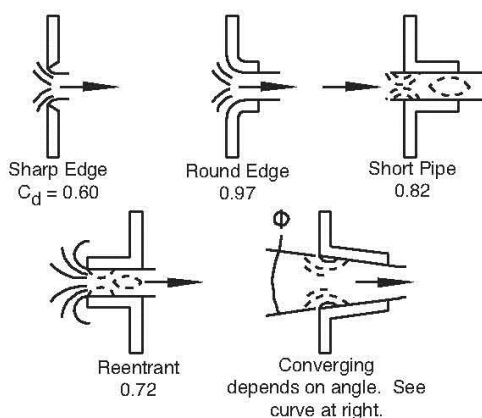
# Appendix

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A.....	Orifice Calculations
B.....	Combustion Constants
C.....	UEL and LEL for Combustibles
D.....	UEL and LEL for Toxic Gases

# CHAPTER 1 – ORIFICES & FLOWS

## COEFFICIENTS OF DISCHARGE FOR VARIOUS TYPES OF ORIFICES



## ORIFICE FLOW FORMULAS

The flow of air or gas through an orifice can be determined by the formula

$$Q = 1658.5 \times A \times C_d \sqrt{\frac{h}{g}}$$

where  $Q$  = flow, cfh

$A$  = area of the orifice, sq. in. (see Pages 57 & 58)

$C_d$  = discharge coefficient of the orifice (see above)

$h$  = pressure drop across the orifice, " w.c.

$g$  = specific gravity of the gas, based on standard air at 1.0 (see Pages 19, 20, & 22 thru 24.)

### 1. Sizing Orifice Plates

To calculate the size of an orifice plate, this equation can be rearranged as follows:

$$A = \frac{Q}{1658.5 \times C_d} \times \sqrt{\frac{g}{h}}$$

### 2. Effect of Changes in Operating Conditions on Flow through an Orifice – General Relationship

$$\frac{Q_2}{Q_1} = \frac{A_2}{A_1} \times \frac{C_{d2}}{C_{d1}} \times \sqrt{\frac{h_2}{h_1}} \times \sqrt{\frac{g_1}{g_2}}$$

If any of the factors in this relationship remain constant from Condition 1 to Condition 2, they can be dropped out of the equation, yielding these simplified relationships. Each of them assumes only one factor has been changed.

#### 2a. Flow Change vs. Orifice Area Change

$$\frac{Q_2}{Q_1} = \frac{A_2}{A_1}$$

#### 2b. Flow Change vs. Pressure Drop Change

$$\frac{Q_2}{Q_1} = \sqrt{\frac{h_2}{h_1}}$$

This is the so-called "square root law."

#### 2c. Flow Change vs. Specific Gravity Change

$$\frac{Q_2}{Q_1} = \sqrt{\frac{g_1}{g_2}}$$

### 3. Effect of Changes in Operating Conditions on Pressure Drop Across an Orifice—General Relationship:

$$\frac{h_2}{h_1} = \left(\frac{Q_2}{Q_1}\right)^2 \times \left(\frac{A_1}{A_2}\right)^2 \times \left(\frac{C_{d1}}{C_{d2}}\right)^2 \times \frac{g_2}{g_1}$$

Again, if any of the factors in this equation are unchanged from Condition 1 to Condition 2, they can be dropped out to form simplified relationships:

#### 3a. Pressure Drop Change vs. Flow Change

$$\frac{h_2}{h_1} = \left(\frac{Q_2}{Q_1}\right)^2$$

This is the square root law, stated another way.

#### 3b. Pressure Drop Change vs. Orifice Area Change

$$\frac{h_2}{h_1} = \left(\frac{A_1}{A_2}\right)^2$$

#### 3c. Pressure Drop Change vs. Specific Gravity Change

$$\frac{h_2}{h_1} = \frac{g_2}{g_1}$$

This relationship may not apply where specific gravity has been changed by a change in gas temperature. See Page 25.

### 4. Effect of Changes in Gas Temperature on Flow and Pressure Drop through an Orifice

Raising a gas's temperature has two effects - it increases the volume and decreases the specific gravity, both in proportion to the ratio of the absolute temperatures. If we are concerned with changes in flows (cfh), these relationships must be used (assuming constant mass flow).

#### 4a. Flow Change vs. Temperature Change

$$\frac{Q_2}{Q_1} = \sqrt{\frac{T_{ABS2}}{T_{ABS1}}} \text{ with } T_{ABS} = \text{absolute temp.}^*$$

\*See page 68 for calculation of the absolute temperature.

#### 4b. Pressure Drop Change vs. Temperature Change

$$\frac{h_2}{h_1} = \frac{T_{ABS2}}{T_{ABS1}}$$

## ORIFICE CAPACITY TABLES LOW PRESSURE GAS

Flows in these tables are based on an orifice pressure drop of 1" w.c. and a coefficient of discharge ( $C_d$ ) of 1.0.

**To determine flow through an orifice of a known diameter:**

1. Locate the orifice diameter in the left-hand column of the table.
2. Read across to the column corresponding to the gas being measured. This is the uncorrected flow.
3. Multiply this flow by the coefficient of discharge of the orifice. (see page 4)
4. Correct this flow to the pressure drop actually measured, using the square root law (equation 2b, page 4).

**Example:** What is the flow of natural gas through a 7/32" diameter sharp edge orifice at 6" w.c. pressure drop?

From the table, uncorrected natural gas flow through a 7/32" orifice is 80.7 cfh at 1" w.c.

$C_d$  for a sharp edge orifice is 0.60 (page 1.1), so corrected flow is  $80.7 \times 0.60 = 48.4$  cfh at 1" w.c. pressure drop.

Per equation 2b, page 4,

$$\frac{Q_2}{Q_1} = \sqrt{\frac{h_2}{h_1}} \text{ or } Q_2 = Q_1 \times \sqrt{\frac{h_2}{h_1}}$$

Substituting the numbers for this case:

$$Q_2 = 48.4 \times \sqrt{\frac{6'' \text{ w.c.}}{1'' \text{ w.c.}}} = 119 \text{ cfh}$$

**To determine the orifice size to handle a known flow at a specified pressure drop, reverse the process:**

1. Correct the known flow to a pressure drop of 1" w.c., using the square root law.
2. Divide the flow by the orifice coefficient.
3. In the orifice table, locate the column for the gas under consideration. In this column, locate the flow closest to the corrected value found in step 2.
4. Read to the left to find the corrected orifice size.

Example: Size a gas jet for a mixer. Entrance to the jet orifice

converges at a 15° included angle. Gas is propane. Required flow is 120 cfh at 30" w.c. pressure drop.

Per equation 2b, page 4,

$$\frac{Q_2}{Q_1} = \sqrt{\frac{h_2}{h_1}} \text{, or } Q_2 = Q_1 \times \sqrt{\frac{h_2}{h_1}}$$

Substituting the numbers for this case:

$$Q_2 = 120 \times \sqrt{\frac{1}{30}} = 22 \text{ cfh}$$

From page 1.1,  $C_d$  for a 15° convergent nozzle is 0.94, so corrected flow is

$$22 \div 0.94 = 23.4 \text{ cfh.}$$

Locate 23.4 cfh in the propane column of the orifice table and then read to the left to find a #26 drill size orifice.

### CAPACITY, CFH @ 1" W.C. PRESSURE DROP AND COEFFICIENT OF DISCHARGE OF 1.0

Drill Size	Dia. In.	Area	Natural Gas		Propane/Air		Butane	
			0.60 Sp. Gr.	1.0 Sp. Gr.	1.29 Sp. Gr.	1.5 Sp. Gr.	2.0 Sp. Gr.	
80	.0135	.000143	.308	.239	.210	.195	.169	
79	.0145	.000165	.355	.275	.242	.225	.195	
1/64	.0156	.00019	.409	.317	.279	.259	.224	
78	.016	.00020	.431	.334	.294	.272	.236	
77	.018	.00025	.538	.417	.367	.340	.295	
76	.020	.00031	.668	.517	.455	.422	.366	
75	.021	.00035	.754	.584	.514	.477	.413	
74	.0225	.00040	.861	.668	.587	.545	.472	
73	.024	.00045	.969	.751	.661	.613	.531	
72	.025	.00049	1.06	.817	.720	.667	.578	
71	.026	.00053	1.14	.884	.778	.722	.625	
70	.028	.00062	1.33	1.03	.910	.844	.731	
69	.0292	.00067	1.44	1.12	.984	.912	.790	
68	.030	.00075	1.61	1.25	1.10	1.02	.885	
1/32	.0312	.00076	1.64	1.27	1.12	1.04	.896	
67	.032	.00080	1.72	1.33	1.17	1.09	.944	
66	.033	.00086	1.85	1.43	1.26	1.17	1.01	
65	.035	.00092	2.07	1.60	1.41	1.31	1.13	
64	.036	.00102	2.20	1.70	1.50	1.39	1.20	
63	.037	.00108	2.33	1.80	1.59	1.47	1.27	
62	.038	.00113	2.43	1.88	1.66	1.54	1.33	
61	.039	.00119	2.56	1.98	1.75	1.62	1.40	
60	.040	.00126	2.71	2.10	1.85	1.72	1.49	
59	.041	.00132	2.84	2.20	1.94	1.8	1.56	
58	.042	.00138	2.97	2.30	2.03	1.88	1.63	

**CAPACITY, CFH @ 1" W.C. PRESSURE DROP  
AND COEFFICIENT OF DISCHARGE OF 1.0**

Drill Size	Dia. In.	Area	Natural Gas		Propane/ Air		Propane 1.5 Sp. Gr.	Butane 2.0 Sp. Gr.
			0.60 Sp. Gr.	1.0 Sp. Gr.	1.29 Sp. Gr.			
57	.043	.00145	3.12	2.42	2.13	1.97	1.71	
56	.0465	.00170	3.66	2.84	2.5	2.32	2.01	
3/64	.0469	.00173	3.73	2.89	2.54	2.36	2.04	
55	.0520	.00210	4.52	3.50	3.08	2.86	2.48	
54	.0550	.0023	4.95	3.84	3.38	3.13	2.71	
53	.0595	.0028	6.03	4.67	4.11	3.81	3.30	
1/16	.0625	.0031	6.68	5.17	4.55	4.22	3.66	
52	.0635	.0032	6.89	5.34	4.7	4.36	3.77	
51	.0670	.0035	7.54	5.84	5.14	4.77	4.13	
50	.070	.0038	8.18	6.34	5.58	5.18	4.48	
49	.073	.0042	9.04	7.01	6.17	5.72	4.95	
48	.076	.0043	9.26	7.17	6.31	5.86	5.07	
5/64	.0781	.0048	10.3	8.01	7.05	6.54	5.66	
47	.0785	.0049	10.5	8.17	7.2	6.67	5.78	
46	.081	.0051	11.	8.51	7.49	6.95	6.02	
45	.082	.0053	11.4	8.84	7.78	7.22	6.25	
44	.086	.0058	12.5	9.67	8.52	7.9	6.84	
43	.089	.0062	13.4	10.3	9.11	8.44	7.31	
42	.0935	.00687	14.8	11.4	10.	9.36	8.1	
3/32	.0937	.0069	14.9	11.5	10.1	9.40	8.14	
41	.096	.0072	15.5	12.	10.6	9.81	8.49	
40	.098	.0075	16.2	12.5	11.	10.2	8.85	
39	.0995	.0078	16.8	13.	11.5	10.6	9.2	
38	.1015	.0081	17.4	13.5	11.9	11.0	9.55	
37	.104	.0085	18.3	14.2	12.5	11.6	10.	
36	.1065	.0090	19.4	15.	13.2	12.3	10.6	
7/64	.1093	.0094	20.2	15.7	13.8	12.8	11.1	
35	.110	.0095	20.5	15.8	14.	12.9	11.2	
34	.111	.0097	20.9	16.2	14.2	13.2	11.4	
33	.113	.0100	21.5	16.7	14.7	13.6	11.8	
32	.116	.0106	22.8	17.7	15.6	14.4	12.5	
31	.120	.0113	24.3	18.8	16.6	15.4	13.3	
1/8	.125	.0123	26.4	20.4	18.	16.7	14.5	
30	.1285	.0130	27.9	21.6	19.	17.6	15.3	
29	.136	.0145	31.1	24.1	21.2	19.7	17.	
28	.1405	.0155	33.3	25.8	22.7	21.	18.2	
9/64	.1406	.0156	33.5	25.9	22.8	21.2	18.3	
27	.144	.0163	35.	27.1	23.9	22.1	19.2	
26	.147	.0174	37.3	28.9	25.5	23.6	20.4	
25	.1495	.0175	37.5	29.1	25.6	23.7	20.6	
24	.152	.0181	38.8	30.1	26.5	24.6	21.3	
23	.154	.0186	39.9	30.9	27.2	25.2	21.9	
5/32	.1562	.0192	41.2	31.9	28.1	26.1	22.6	
22	.157	.0193	41.4	32.1	28.2	26.2	22.7	
21	.159	.0198	42.5	32.9	29.	26.9	23.3	
20	.161	.0203	43.6	33.7	29.7	27.5	23.9	
19	.166	.0216	46.3	35.9	31.6	29.3	25.4	
18	.1695	.0226	48.5	37.6	33.1	30.7	26.6	
11/64	.1719	.0232	49.8	38.6	33.9	31.5	27.3	
17	.175	.0235	50.4	39.1	34.4	31.9	27.6	
16	.177	.0246	52.8	40.9	36.	33.4	28.9	
15	.180	.0254	54.5	42.2	37.2	34.5	29.9	
14	.182	.0260	55.8	43.2	38.	35.3	30.6	
13	.185	.0269	57.7	44.7	39.4	36.5	31.6	
3/16	.1875	.0276	59.2	45.9	40.4	37.5	32.4	



**CAPACITY, CFH @ 1" W.C. PRESSURE DROP  
AND COEFFICIENT OF DISCHARGE OF 1.0**

Drill Size	Dia. In.	Area	Natural Gas	Air	Propane/ Air	Propane	Butane
			0.60 Sp. Gr.	1.0 Sp. Gr.	1.29 Sp. Gr.	1.5 Sp. Gr.	2.0 Sp. Gr.
12	.189	.02805	60.2	46.6	41.	38.1	33.
11	.191	.02865	61.5	47.6	41.9	38.9	33.7
10	.1935	.0294	63.1	48.9	43.	39.9	34.6
9	.196	.0302	64.8	50.2	44.2	41.	35.5
8	.199	.0311	66.7	51.7	45.5	42.2	36.5
7	.201	.0316	67.8	52.5	46.2	42.9	37.1
13/64	.2031	.0324	69.5	53.8	47.4	44.	38.1
6	.204	.0327	70.2	54.3	47.8	44.4	38.4
5	.2055	.0332	71.2	55.2	48.6	45.1	39.
4	.209	.0343	73.6	57.0	50.2	46.5	40.3
3	.213	.0356	76.4	59.2	52.1	48.3	41.8
7/32	.2187	.0376	80.7	62.5	55.	51.	44.2
2	.221	.0384	82.4	63.8	56.2	52.1	45.1
1	.228	.0409	87.8	68.	59.8	55.5	48.1
A	.234	.0430	92.3	71.5	62.9	58.4	50.5
15/64	.2343	.0431	92.5	71.6	63.1	58.5	50.7
B	.238	.0444	95.3	73.8	65.	60.3	52.2
C	.242	.0460	98.7	76.5	67.3	62.4	54.1
D	.246	.0475	102.	78.9	69.5	64.5	55.8
1/4	.250	.0491	105.	81.6	71.8	66.6	57.7
F	.257	.0519	111.	86.3	75.9	70.4	61.
G	.261	.0535	115.	88.9	78.3	72.6	62.9
17/64	.2656	.0554	119.	92.1	81.1	75.2	65.1
H	.266	.0556	119.3	92.4	81.4	75.4	65.3
I	.272	.0580	124.	96.4	84.9	78.7	68.2
J	.277	.0601	129.	99.9	87.9	81.6	70.6
K	.281	.0620	133.	103.	90.7	84.1	72.9
9/32	.2812	.0621	133.2	103.2	90.9	84.3	73.
L	.290	.0660	142.	110.	96.6	89.6	77.6
M	.295	.0683	147.	113.	99.9	92.7	80.3
19/64	.2968	.0692	148.	115.	101.	93.9	81.3
N	.302	.0716	154.	119.	105.	97.2	84.1
5/16	.3125	.0767	165.	127.	112.	104.	90.1
O	.316	.0784	168.	130.	115.	106.	92.1
P	.323	.0820	176.	136.	120.	111.	96.4
21/64	.3281	.0846	182.	141.	124.	115.	99.4
Q	.332	.0866	186.	144.	127.	118.	102.
R	.339	.0901	193.	150.	132.	122.	106.
11/32	.3437	.0928	199.	154.	136.	126.	109.
S	.348	.0950	204.	158.	139.	129.	112.
T	.358	.1005	216.	167.	147.	136.	118.
23/64	.3593	.1014	218.	169.	148.	138.	119.
U	.368	.1063	228.	177.	156.	144.	125.
3/8	.375	.1104	237.	184.	162.	150.	130.
V	.377	.1116	239.	185.	163.	151.	131.
W	.386	.1170	251.	194.	171.	159.	137.
25/64	.3906	.1198	257.	199.	175.	163.	141.
X	.397	.1236	265.	205.	181.	168.	145.
Y	.404	.1278	274.	212.	187.	173.	150.
13/32	.4062	.1296	278.	215.	190.	176.	152.
Z	.413	.1340	288.	223.	196.	182.	157.
27/64	.4219	.1398	300.	232.	205.	190.	164.
7/16	.4375	.1503	322.	250.	220.	204.	177.
29/64	.4531	.1613	346.	268.	236.	219.	190.
15/32	.4687	.1726	370.	287.	253.	234.	203.

**CAPACITY, CFH @ 1" W.C. PRESSURE DROP  
AND COEFFICIENT OF DISCHARGE OF 1.0**

Drill Size	Dia. In.	Area	Natural Gas		Propane/ Air		Butane	
			0.60 Sp. Gr.	1.0 Sp. Gr.	1.29 Sp. Gr.	1.5 Sp. Gr.	2.0 Sp. Gr.	
31/64	.4843	.1843	395.	306.	270.	250.	217.	
1/2	.50	.1963	421.	326.	287.	266.	231.	
33/64	.5156	.2088	448.	347.	306.	283.	245.	
17/32	.5312	.2217	476.	368.	324.	301.	261.	
35/64	.5468	.2349	504.	390.	344.	319.	276.	
9/16	.5625	.2485	533.	413.	364.	337.	292.	
37/64	.5781	.2625	563.	436.	384.	356.	308.	
19/32	.5937	.2769	594.	460.	405.	376.	325.	
39/64	.6093	.2916	626.	485.	427.	396.	343.	
5/8	.625	.3068	658.	510.	449.	416.	361.	
41/64	.6406	.3223	691.	536.	472.	437.	379.	
21/32	.6562	.3382	725.	562.	495.	459.	397.	
43/64	.6718	.3545	760.	589.	519.	481.	417.	
11/16	.6875	.3712	796.	617.	543.	504.	436.	
45/64	.7031	.3883	833.	645.	568.	527.	456.	
23/32	.7187	.4057	870.	674.	594.	551.	477.	
47/64	.7343	.4236	909.	704.	620.	575.	498.	
3/4	.750	.44179	948.	734.	646.	599.	519.	
49/64	.7656	.46040	988.	765.	674.	625.	541.	
25/32	.7813	.47937	1029.	796.	701.	651.	563.	
51/64	.7969	.49873	1070.	829.	730.	677.	586.	
13/16	.8125	.51849	1112.	862.	759.	704.	609.	
53/64	.8281	.53862	1156.	895.	788.	731.	633.	
27/32	.8438	.55914	1200.	929.	818.	759.	657.	
55/64	.8594	.5800	1244.	964.	849.	787.	682.	
7/8	.8750	.60132	1290.	999.	880.	816.	707.	
29/32	.9062	.64504	1384.	1072.	944.	875.	758.	
15/16	.9375	.69029	1481.	1147.	1010.	937.	811.	
31/32	.9688	.73708	1581.	1225.	1079.	1000.	866.	
1	1.0	.7854	1685.	1305.	1149.	1066.	923.	
1-1/16	1.063	.88664	1902.	1474.	1297.	1203.	1042.	
1-1/8	1.125	.99402	2133.	1652.	1455.	1349.	1168.	
1-3/16	1.188	1.1075	2376.	1841.	1621.	1503.	1302.	
1-1/4	1.250	1.2272	2633.	2040.	1796.	1665.	1442.	
1-5/16	1.313	1.3530	2903.	2249.	1980.	1836.	1590.	
1-3/8	1.375	1.4849	3186.	2468.	2173.	2015.	1745.	
1-1/2	1.5	1.7671	3791.	2937.	2586.	2398.	2077.	
1-9/16	1.563	1.9174	4114.	3187.	2806.	2602.	2253.	
1-5/8	1.625	2.0739	4450.	3447.	3035.	2814.	2437.	
1-11/16	1.688	2.2365	4799.	3717.	3273.	3035.	2628.	
1-3/4	1.75	2.4053	5161.	3998.	3520.	3264.	2827.	
1-13/16	1.813	2.5802	5536.	4288.	3776.	3501.	3032.	
1-7/8	1.875	2.7612	5924.	4589.	4040.	3747.	3245.	
1-15/16	1.938	2.9498	6329.	4903.	4316.	4003.	3467.	
2	2.0	3.1416	6741.	5221.	4597.	4263.	3692.	
2-1/8	2.125	3.5466	7610.	5894.	5190.	4813.	4168.	
2-1/4	2.250	3.9761	8531.	6608.	5818.	5396.	4673.	
2-3/8	2.375	4.4301	9505.	7363.	6483.	6012.	5206.	
2-1/2	2.50	4.9087	10532.	8158.	7183.	6661.	5769.	
2-5/8	2.625	5.4119	11612.	8995.	7919.	7344.	6360.	
2-3/4	2.75	5.9396	12744.	9872.	8691.	8060.	6980.	
2-7/8	2.875	6.4918	13929.	10789.	9499.	8809.	7629.	

**Table 1 — Combustion Constants — Reference 1**

No.	Substance	Formula	Molecular Weight <sup>a</sup>	Density <sup>b</sup> lb per ft <sup>3</sup>	Specific Volume <sup>b</sup> ft <sup>3</sup> per lb	Specific Gravity <sup>b</sup> (air=1)	Heat of Combustion <sup>c</sup>			ft <sup>3</sup> per ft <sup>3</sup> of Combustible			lb per lb of Combustible			Theor air lb/10,000 Btu							
							Gross <sup>d</sup>	Net <sup>e</sup>	Btu per lb	O <sub>2</sub>	N <sub>2a</sub>	Air	Flue Products CO <sub>2</sub> H <sub>2</sub> O N <sub>2a</sub>	O <sub>2</sub>	N <sub>2a</sub>		Air	Flue Gas Products CO <sub>2</sub> H <sub>2</sub> O N <sub>2a</sub>					
1	Carbon	C	12.0110	—	—	—	14,093	14,093	—	1.0	3,773	4,773	1.0	—	3,773	2,664	8,846	11,510	3,664	—	8,846	8,167	
2	Hydrogen	H <sub>2</sub>	2.0159	0.0053	188,245	0.0696	324.2	273.9	61,029	51,558	0.5	1,887	2,387	1.0	1.0	1,887	7,936	26,353	34,290	—	8,937	26,353	5,619
3	Oxygen	O <sub>2</sub>	31.9988	0.0844	11,850	1.1053	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—
4	Nitrogen	N <sub>2</sub>	28.0134	0.0738	13,543	0.9671	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—
4	Nitrogen (atm.) <sup>f</sup>	N <sub>2a</sub>	28.1580	0.0742	13,474	0.9720	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—
5	Carbon monoxide	CO	28.0104	0.0738	13,542	0.9672	320.6	320.6	4,342	4,342	0.5	1,887	2,387	1.0	—	1,887	0,571	1,897	2,468	1,571	—	1,897	5,684
6	Carbon dioxide	CO <sub>2</sub>	44.0098	0.1166	8,574	1.5277	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—
<b>Paraffin series C<sub>n</sub>H<sub>2n+2</sub></b>																							
7	Methane	CH <sub>4</sub>	16.0428	0.0424	23,608	0.5548	1012	911	23,891	21,511	2.0	7,547	9,547	1.0	2.0	7,547	3,989	13,246	17,235	2,743	2,246	13,246	7,214
8	Ethane	C <sub>2</sub> H <sub>6</sub>	30.0697	0.0799	12,514	1.0466	1785	1634	22,334	20,429	3.5	13,206	16,706	2.0	3.0	13,206	3,724	12,367	16,092	2,927	1,797	12,367	7,205
9	Propane	C <sub>3</sub> H <sub>8</sub>	44.0966	0.1183	8,456	1.5489	2561	2359	21,653	19,921	5.0	18,866	23,866	3.0	4.0	18,866	3,628	12,047	15,676	2,994	1,634	12,047	7,239
10	n-Butane	C <sub>4</sub> H <sub>10</sub>	58.1235	0.1585	6,310	2.0758	3376	3124	21,299	19,657	6.5	24,526	31,026	4.0	5.0	24,526	3,578	11,882	15,460	3,029	1,550	11,882	7,259
11	Isobutane	C <sub>4</sub> H <sub>10</sub>	58.1235	0.1580	6,328	2.0699	3355	3104	21,231	19,589	6.5	24,526	31,026	4.0	5.0	24,526	3,578	11,882	15,460	3,029	1,550	11,882	7,282
12	n-Pentane	C <sub>5</sub> H <sub>12</sub>	72.1504	0.2019	4,952	2.6450	4258	3956	21,085	19,498	8.0	30,186	38,186	5.0	6.0	30,186	3,548	11,781	15,329	3,050	1,498	11,781	7,270
13	Isopentane	C <sub>5</sub> H <sub>12</sub>	72.1504	0.2001	4,999	2.6202	4210	3908	21,043	19,455	8.0	30,186	38,186	5.0	6.0	30,186	3,548	11,781	15,329	3,050	1,498	11,781	7,284
14	Neopentane	C <sub>5</sub> H <sub>12</sub>	72.1504	0.1984 <sup>g</sup>	5,040 <sup>g</sup>	2.5989 <sup>g</sup>	4159 <sup>g</sup>	3857	20,958 <sup>g</sup>	19,370	8.0	30,186	38,186	5.0	6.0	30,186	3,548	11,781	15,329	3,050	1,498	11,781	7,314
15	n-Hexane	C <sub>6</sub> H <sub>14</sub>	86.1773	0.2508	3,987	3.2849	5252	4900	20,943	19,392	9.5	35,846	45,346	6.0	7.0	35,846	3,527	11,713	15,240	3,064	1,463	11,713	7,277
<b>Olefin series C<sub>n</sub>H<sub>2n</sub></b>																							
16	Ethylene	C <sub>2</sub> H <sub>4</sub>	28.0538	0.0744	13,447	0.9740	1609	1509	21,643	20,282	3.0	11,320	14,320	2.0	2.0	11,320	3,422	11,362	14,784	3,138	1,284	11,362	6,831
17	Propylene	C <sub>3</sub> H <sub>6</sub>	42.0807	0.1127	8,874	1.4760	2371	2220	21,039	19,678	4.5	16,980	21,480	3.0	3.0	16,980	3,422	11,362	14,784	3,138	1,284	11,362	7,027
18	n-Butene (Butylene)	C <sub>4</sub> H <sub>8</sub>	56.1076	0.1524 <sup>h</sup>	6,560 <sup>h</sup>	1.9966 <sup>h</sup>	3175 <sup>h</sup>	2974	20,831 <sup>h</sup>	19,470	6.0	22,640	28,640	4.0	4.0	22,640	3,422	11,362	14,784	3,138	1,284	11,362	7,097
19	Isobutene	C <sub>4</sub> H <sub>8</sub>	56.1076	0.1524 <sup>h</sup>	6,561 <sup>h</sup>	1.9964 <sup>h</sup>	3156 <sup>h</sup>	2955	20,704 <sup>h</sup>	19,343	6.0	22,640	28,640	4.0	4.0	22,640	3,422	11,362	14,784	3,138	1,284	11,362	7,141
20	n-Pentene	C <sub>5</sub> H <sub>10</sub>	70.1345	0.1947 <sup>h</sup>	5,135 <sup>h</sup>	2.5508 <sup>h</sup>	4032 <sup>h</sup>	3781	20,704 <sup>h</sup>	19,343	7.5	28,300	35,800	5.0	5.0	28,300	3,422	11,362	14,784	3,138	1,284	11,362	7,140
<b>Aromatic series C<sub>n</sub>H<sub>n-6</sub></b>																							
21	Benzene	C <sub>6</sub> H <sub>6</sub>	78.1137	0.2213	4,518	2.8989	4024	3873	18,179	17,446	7.5	28,300	35,800	6.0	3.0	28,300	3,072	10,201	13,274	3,380	0,692	10,201	7,302
22	Toluene	C <sub>7</sub> H <sub>8</sub>	92.1406	0.2750 <sup>h</sup>	3,637 <sup>h</sup>	3.6016 <sup>h</sup>	5068 <sup>h</sup>	4867	18,430 <sup>h</sup>	17,602	9.0	33,959	42,959	7.0	4.0	33,959	3,125	10,378	13,504	3,343	0,782	10,378	7,327
23	Xylene	C <sub>8</sub> H <sub>10</sub>	106.1675	0.3480 <sup>h</sup>	2,874 <sup>h</sup>	4.5576 <sup>h</sup>	6480 <sup>h</sup>	6228	18,622 <sup>h</sup>	17,723	10.5	39,619	50,119	8.0	5.0	39,619	3,164	10,508	13,673	3,316	0,848	10,508	7,342
<b>Miscellaneous</b>																							
24	Acetylene	C <sub>2</sub> H <sub>2</sub>	26.0379	0.0691	14,480	0.9046	1484	1433	21,482	20,749	2.5	9,433	11,933	2.0	1.0	9,433	3,072	10,201	13,274	3,380	0,692	10,201	6,179
25	Naphthalene	C <sub>10</sub> H <sub>8</sub>	128.1736	0.3384 <sup>h</sup>	2,955 <sup>h</sup>	4.4323 <sup>h</sup>	5866	5665	17,335	16,739	12.0	45,279	57,279	10.0	4.0	45,279	2,995	9,947	12,943	3,434	0,562	9,947	7,467
26	Methyl alcohol	CH <sub>3</sub> OH	32.0422	0.0846 <sup>h</sup>	11,820 <sup>h</sup>	1.1081 <sup>h</sup>	868 <sup>h</sup>	768	10,265 <sup>h</sup>	9,073	1.5	5,660	7,160	1.0	2.0	5,660	1,498	4,974	6,472	1,373	1,124	4,974	6,305
27	Ethyl alcohol	C <sub>2</sub> H <sub>5</sub> OH	46.0691	0.1216 <sup>h</sup>	8,224 <sup>h</sup>	1.5927 <sup>h</sup>	1602 <sup>h</sup>	1451	13,172 <sup>h</sup>	11,929	3.0	11,320	14,320	2.0	3.0	11,320	2,084	6,919	9,003	1,911	1,173	6,919	6,835
28	Ammonia	NH <sub>3</sub>	17.0306	0.0454 <sup>a</sup>	22,008 <sup>a</sup>	0.5951 <sup>a</sup>	440 <sup>a</sup>	364	9680 <sup>a</sup>	7998	0.75	2,830	3,580	—	1.5	3,330	1,409	4,679	6,088	—	1,587	5,502	6,290
29	Sulfur	S	32.0660	—	—	—	—	—	3980	3980	1.0	3,773	4,773	1.0	—	3,773	1,000	3,320	4,310	1,998	—	3,320	10,829
30	Hydrogen sulfide	H <sub>2</sub> S	34.0819	0.0907	11,030	1.1875	643	593	7094	6534	1.5	5,660	7,160	1.0	1.0	5,660	1,410	4,682	6,093	1,880	0,529	4,682	8,576
31	Sulfur dioxide	SO <sub>2</sub>	64.0648	0.1722 <sup>a</sup>	5,806 <sup>a</sup>	2.2558 <sup>a</sup>	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—
32	Water vapor	H <sub>2</sub> O	18.0153	0.0503	19,863	0.6594	50.312	0.0	1059.8	0.0	—	—	—	—	—	—	—	—	—	—	—	—	—
33	Air <sup>d</sup>	—	28.9625	0.0763	13,098	1.0000	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—

All gas volumes corrected to 60°F and 14.696 psi dry.  
a 1987 Atomic Weights: C=12.011, H=1.00794, O=15.9994, N=14.0067, S=32.066.  
b Densities calculated from ideal values and compressibility factor given in ASTM D3588-98. Some of the materials cannot exist as gases at 60°F and 14.696 psi, in which case the values are theoretical ones. Under the actual concentrations in which these materials are present, their partial pressure is low enough to keep them as gases.  
c For gases saturated with water at 60°F and 14.696 psi, 1.74% of the Btu value must be deducted, Reference 2.  
d Reference 2, ASTM 3588-98.  
e Correction from gross to net heating value determined by deducting the HV shown for water vapor times the moles of H<sub>2</sub>.  
f Reference 3, Jones, F.E.  
g Gas Processors Suppliers Association (GPSA) Data Book, Fig 23-2, Physical Constants, 1987.  
h Either the density or the compressibility factor has been assumed.



# Combustibles

Material	LEL (%/Vol)	UEL (%/Vol)	TLV/TWA (ppm)	IDLH (ppm)	Density (Air = 1.0)
Acetone	2.5	12.8	750	2,500	2.0
Acetylene	2.5	100.0	-A-	-A-	.9
Ammonia	15.0	28.0	25	300	0.6
Benzene	1.2	7.8	1.0	500	2.6
Butane	1.6	8.4	800	-U-	2.0
n-Butyl Acetate	1.7	7.6	150	1,700	4.0
Diborane	0.8	88.0	0.1	15	1.0
Ethane	3.0	12.5	-A-	-A-	1.0
Ethanol	3.3	19.0	1,000	-U-	1.6
Ethyl Acetate	2.0	11.5	400	2,000	3.0
Ethyl Ether	1.9	36.0	400	1,900	2.6
Ethylene Oxide	3.0	100.0	1	-C-	1.5
Gasoline (100 Octane)	1.4	7.6	300	-U-	3-4.0
Heptane	1.05	6.7	400	750	3.5
Hexane	1.1	7.5	50	1,100	3.0
Hydrogen	4.0	75.0	-A-	-A-	0.1
Isopropyl Alcohol	2.0	12.0	400	2,000	2.1
Methane	5.0	15.0	-A-	-A-	0.6
Methanol	6.0	36.0	200	6,000	1.1
Methyl Ethyl Ketone	1.4	11.4	200	3,000	2.5
Pentane	1.5	7.8	600	15,000	2.5
Propane	2.1	9.5	1,000	2,100	1.6
Propylene Oxide	2.3	36.0	20	400	2.0
Styrene	0.9	6.8	50	700	3.6
Toluene	1.1	7.1	50	500	3.1
Turpentine	0.8	?	100	800	4.7
Vinyl Acetate	2.6	13.4	10	-U-	3.0
Vinyl Chloride	3.6	33.0	1.0	-C-	2.2
Xylene	0.9	6.7	100	900	3.7

- LEL Lower Explosive Limit
- UEL Upper Explosive Limit
- PPM Parts Per Million
- TLV/TWA Threshold Limit Value/Time Weighted Average
- IDLH Immediately Dangerous to Life or Health
- Density < 1.0 = lighter than air  
> 1.0 = heavier than air
- A Asphyxiant
- C Carcinogen
- U Data Not Available

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## Toxics

Material	TLV/TWA (ppm)	IDLH	LEL (ppm)	LEL (%Vol)	Density (Air=1)
Acetone	750	2,500	25,000	2.5	2.0
Ammonia	25	300	160,000	16.0	0.6
Benzene	1.0	-C-	12,000	1.2	2.6
Butane	800	-U-	16,000	1.6	2.0
n-Butyl Acetate	150	1,700	17,000	1.7	4.0
Carbon Dioxide	5,000	40,000	N/C	N/C	1.5
Carbon Monoxide	25	1,200	125,000	12.5	1.0
Chlorine	0.5	10	N/C	N/C	2.5
Ethylene Oxide	1	-C-	30,000	3.0	1.5
Ethyl Ether	400	19,000	19,000	1.9	2.6
Gasoline	300	-U-	14,000	1.4	3-4.0
Heptane	400	750	10,500	1.05	3.5
Hexane	50	1,100	11,000	1.0	3.0
Hydrogen Cyanide	10	50	56,000	5.6	0.9
Hydrogen Sulfide	10	100	40,000	4.0	1.2
Isopropyl Alcohol	400	2,000	20,000	2.0	2.1
Methyl Acetate	200	3,100	31,000	3.1	2.6
Methanol	200	6,000	60,000	6.0	1.1
Methyl Chloride	50	2,000	81,000	8.1	1.8
Methyl Ethyl Ketone	200	3,000	14,000	1.4	2.5
Methyl Methacrylate	100	1,000	17,000	1.7	3.5
Nitric Oxide	25	100	N/C	N/C	1.0
Nitrogen Dioxide	3	20	N/C	N/C	1.6
Pentane	600	15,000	15,000	1.5	2.5
n-Propyl Acetate	200	1,700	17,000	1.7	3.5
Styrene	50	700	9,000	.9	3.6
Sulfur Dioxide	2	100	N/C	N/C	2.2
1,1,1-Trichloroethane	350	700	75,000	7.5	4.6
Toluene	50	500	11,000	1.1	3.2
Trichloroethylene	50	1,000	80,000	8.0	4.5
Turpentine	100	800	8,000	0.8	4.7
Vinyl Chloride	1.0	-C-	36,000	3.6	2.2
Xylene	100	900	9,000	.9	3.7

LEL Lower Explosive Limit UEL Upper Explosive Limit  
 PPM Parts Per Million TLV/TWA Threshold Limit Value/Time Weighted Average  
 IDLH Immediately Dangerous to Life or Health Density < 1.0 = lighter than air > 1.0 = heavier than air  
 C Carcinogen N/C Not Combustible

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## Specific Gravity Correction Factors

Gas Meter is Calibrated With	Gas Being Used																
	Hydrogen	Helium	Methane	Ammonia	Neon	Acetylene	Nitrogen/Carbon Monoxide	Ethylene	Air	Ethane	Oxygen	Hydrogen Sulfide	Argone	Nitrous / Carbon Dioxide	Propane	Butane	Sulfur Dioxide
Hydrogen	1	0.70	0.35	0.34	0.32	0.28	0.27	0.27	0.26	0.26	0.25	0.24	0.22	0.21	0.21	0.18	0.18
Helium	1.41	1	0.50	0.48	0.45	0.38	0.38	0.38	0.37	0.36	0.35	0.34	0.32	0.30	0.30	0.26	0.25
Methane	2.82	2	1	0.97	0.89	0.78	0.76	0.75	0.74	0.73	0.71	0.68	0.63	0.60	0.59	0.52	0.49
Ammonia	2.92	2.06	1.03	1	0.92	0.81	0.78	0.78	0.77	0.75	0.73	0.70	0.66	0.62	0.62	0.54	0.51
Neon	3.17	2.25	1.12	1.06	1	0.88	0.85	0.84	0.83	0.82	0.80	0.76	0.71	0.67	0.67	0.58	0.55
Acetylene	3.62	2.56	1.28	1.24	1.14	1	0.97	0.96	0.95	0.93	0.91	0.87	0.81	0.77	0.76	0.66	0.63
Nitrogen/Carbon Monoxide	3.74	2.64	1.32	1.28	1.18	1.03	1	1	0.98	0.96	0.94	0.90	0.84	0.80	0.79	0.68	0.65
Ethylene	3.74	2.66	1.33	1.26	1.18	1.03	1	1	1.01	0.96	0.94	0.90	0.84	0.80	0.79	0.69	0.66
Air	3.61	2.69	1.35	1.30	1.20	1.04	1.02	1.01	1	0.98	0.95	0.92	0.85	0.81	0.80	0.70	0.66
Ethane	3.90	2.76	1.38	1.33	1.23	1.08	1.04	1.04	1.02	1	0.98	0.94	0.88	0.83	0.82	0.71	0.68
Oxygen	4	2.82	1.41	1.36	1.26	1.10	1.06	1.06	1.05	1.02	1	0.95	0.90	0.85	0.84	0.73	0.70
Hydrogen Sulfide	4.15	2.94	1.47	1.42	1.31	1.15	1.11	1.11	1.09	1.06	1.04	1	0.93	0.88	0.88	0.76	0.72
Argon	4.45	3.15	1.58	1.52	1.40	1.23	1.19	1.18	1.17	1.14	1.12	1.07	1	0.94	0.94	0.82	0.78
Nitrous Oxide / Carbon Dioxide	4.70	3.33	1.67	1.61	1.48	1.30	1.26	1.25	1.24	1.21	1.18	1.13	1.06	1	0.99	0.88	0.82
Propane	4.76	3.36	1.68	1.63	1.50	1.31	1.27	1.26	1.25	1.22	1.19	1.15	1.07	1.01	1	0.87	0.83
Butane	5.46	3.66	1.93	1.67	1.72	1.51	1.46	1.45	1.43	1.40	1.37	1.32	1.22	1.16	1.15	1	0.95
Sulfur Dioxide	5.72	4.05	2.03	1.96	1.81	1.58	1.53	1.52	1.50	1.47	1.43	1.38	1.28	1.22	1.20	1.05	1